



Australian Centre for Advanced Photovoltaics

“Emergence of Perovskite Solar Cells”

Martin A. Green

University of New South Wales (UNSW)

Sydney, Australia



Photovoltaics - Electricity from Sunlight



ACAP

Science and Nature agree:

Big solar news of 2013!

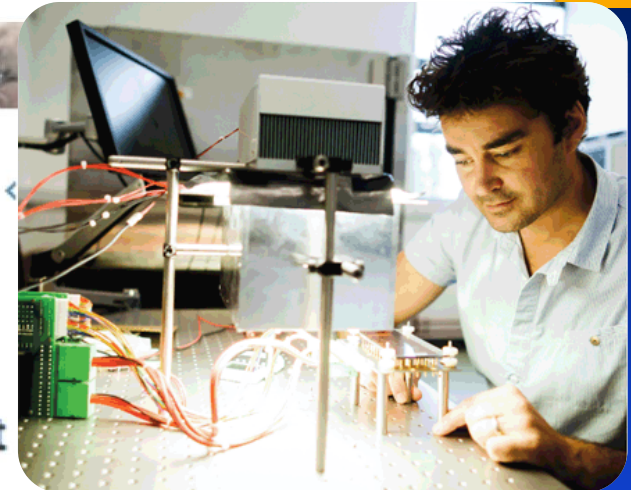
BREAKTHROUGH OF THE YEAR

Science 20 December 2013:
Vol. 342 no. 6165 pp. 1438-1439
DOI: 10.1126/science.342.6165.1438-b

NEWS

Newcomer Juices Up the Race to Harness Sunlight

Up-and-coming solar cell materials called perovskites made such rapid progress this year that the editors of *Science* picked them as a runner-up for Breakthrough of the Year.



HENRY SNAITH: Sun worshipper
An energetic physicist pushes a promising solar-cell material into the spotlight.
By Mark Peplow

365 days: Nature's 10

Ten people who mattered this year.

18 December 2013



Douglas Pyl/Pixanta Photography

"I always wanted to be an inventor," says Henry Snaithe happily. The 35-year-old physicist at the University of Oxford, UK, has fulfilled that childhood ambition in spectacular style.

from Sunlight



18 December 2013



EU PVSEC 2013

28th European PV Solar Energy Conference and Exhibition





Nature photonics review (July 2014)

nature
photonics

REVIEW ARTICLE

PUBLISHED ONLINE: XX JULY 2014 | DOI: 10.1038/NPHOTON.2014.134

The emergence of perovskite solar cells

Martin A. Green¹, Anita Ho-Baillie¹ and Henry J. Snaith²

The past two years have seen the unprecedentedly rapid emergence of a new class of solar cell based on mixed organic–inorganic halide perovskites. Although the first efficient solid-state perovskite cells were reported only in mid-2012, extremely rapid progress was made during 2013 with energy conversion efficiencies reaching a confirmed 16.2% at the end of the year. This increased to a confirmed efficiency of 17.9% in early 2014, with unconfirmed values as high as 19.3% claimed. Moreover, a broad range of different fabrication approaches and device concepts is represented among the highest performing devices — this diversity suggests that performance is still far from fully optimized. This Review briefly outlines notable achievements to date, describes the unique attributes of these perovskites leading to their rapid emergence and discusses challenges facing the successful development and commercialization of perovskite solar cells.



Nature photonics review (July 2014)

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The emergence of perovskite solar cells

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The past two years have seen the unprecedentedly rapid emergence of a new class of solar cells based on organic halide perovskites. Although the first efficient solid-state perovskite cells were reported in 2009, rapid progress was made during 2013 with energy conversion efficiencies reaching a confirmed value of 17.1%. This increased to a confirmed efficiency of 17.9% in early 2014, with unconfirmed values as high as 20.1%. A broad range of different fabrication approaches and device concepts is represented among these reports, and this diversity suggests that performance is still far from fully optimized. This Review briefly updates the field, describes the unique attributes of these perovskites leading to their rapid emergence and discusses the successful development and commercialization of perovskite solar cells.



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Photovoltaics



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An energetic physicist pushes a promising solar-cell material into the spotlight.

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Douglas Fry/Piranha Photography

"I always wanted to be an inventor," says Henry Snaith happily. The 35-year-old physicist at the University of Oxford, UK, has fulfilled that childhood ambition in spectacular style.

Perovskite solar cells

Henry J. Snaith²

The rapid emergence of a new class of solar cells, efficient solid-state perovskite cells were reported with power conversion efficiencies reaching a confirmed value of 22% in early 2014, with unconfirmed values as high as 25%. This Review and device concepts is represented among the most promising materials far from fully optimized. This Review briefly reviews the perovskites leading to their rapid emergence and the challenges of perovskite solar cells.



Photovoltaics



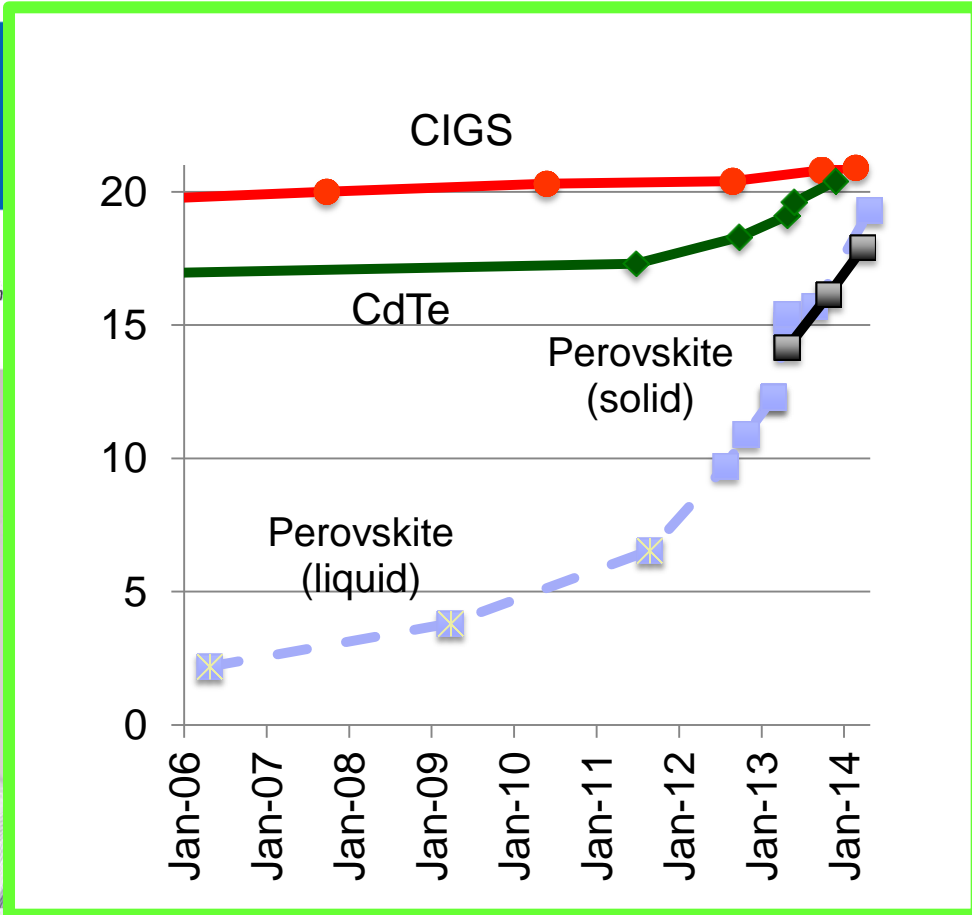
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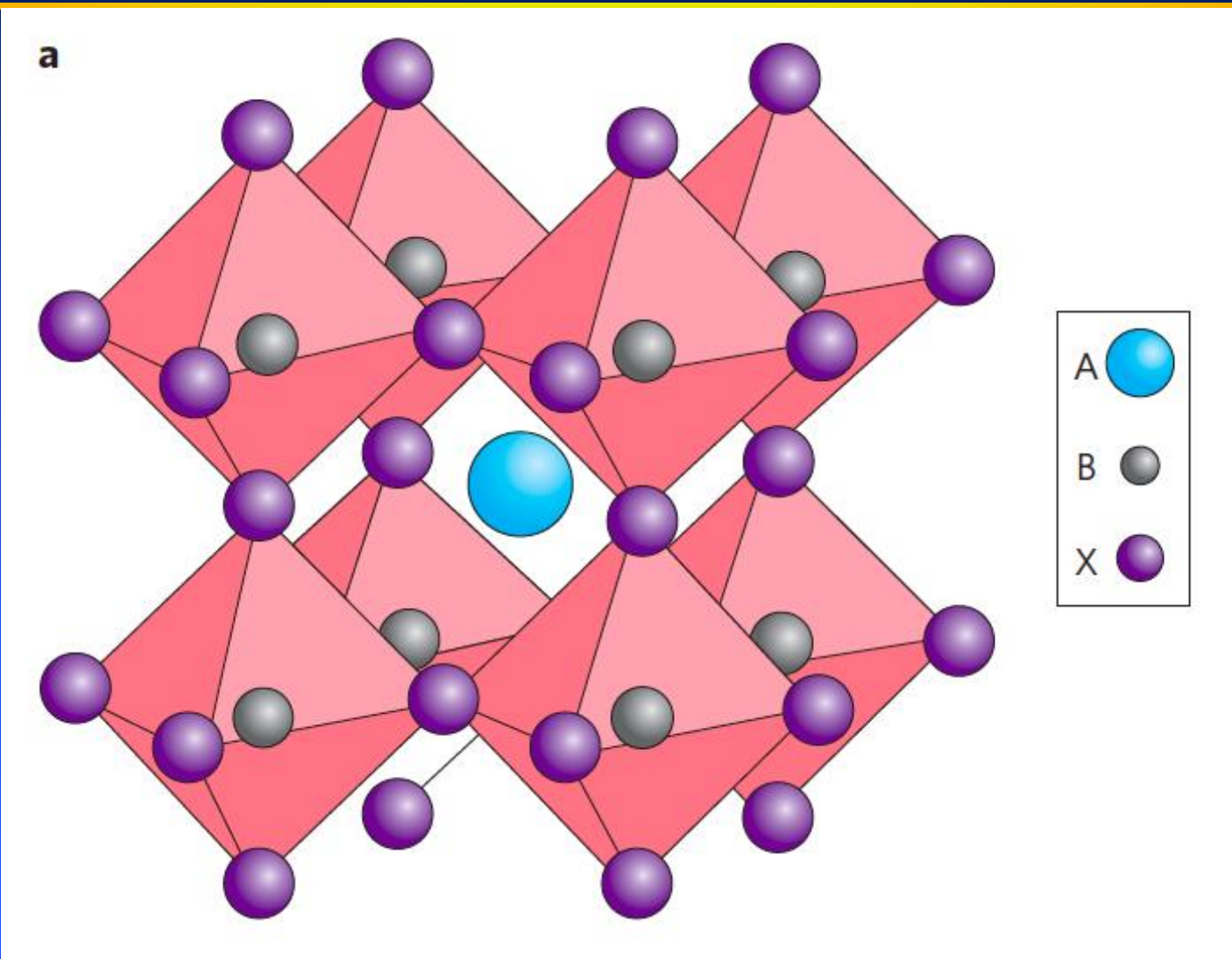


Photovoltaics



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Perovskite



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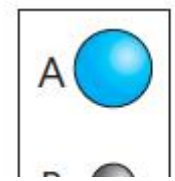
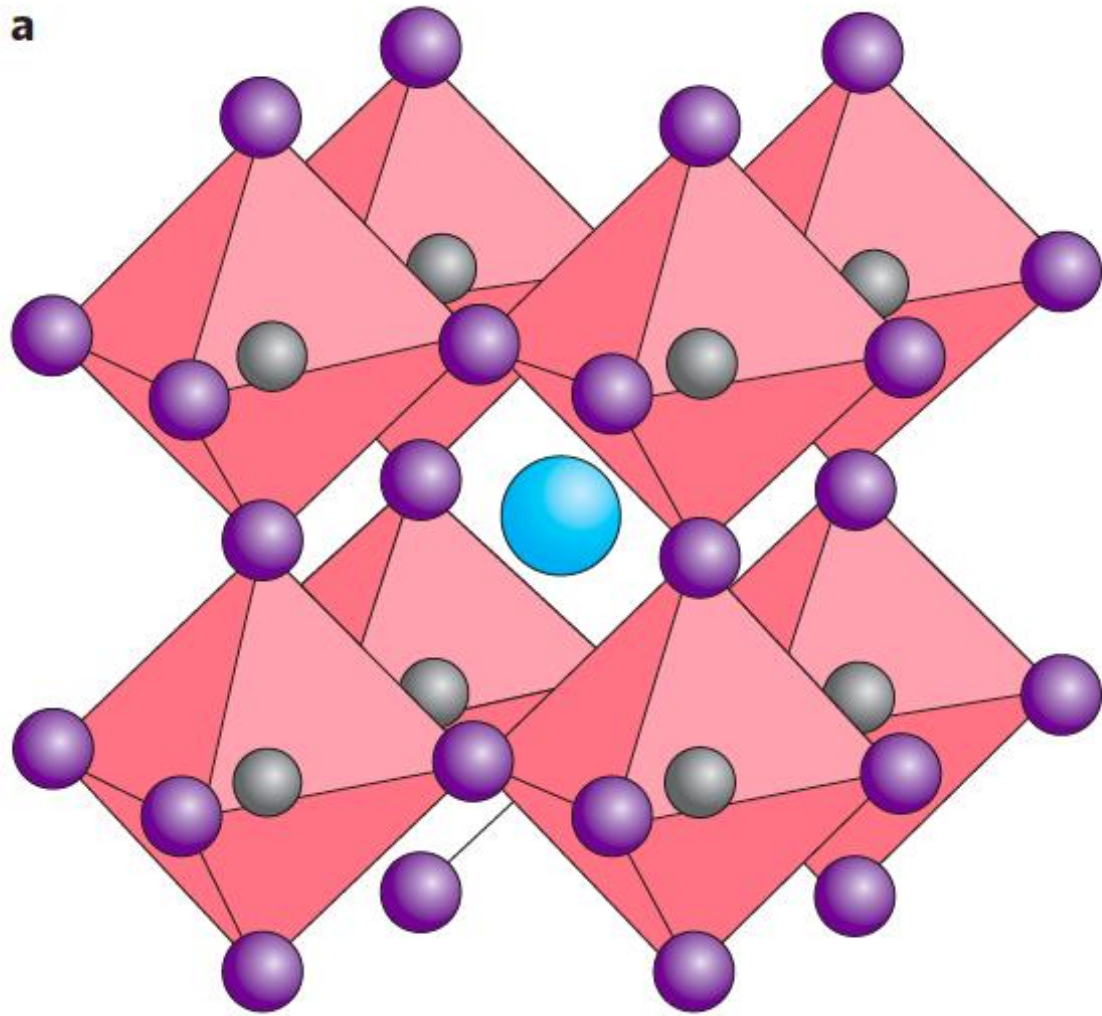
in Sunlight



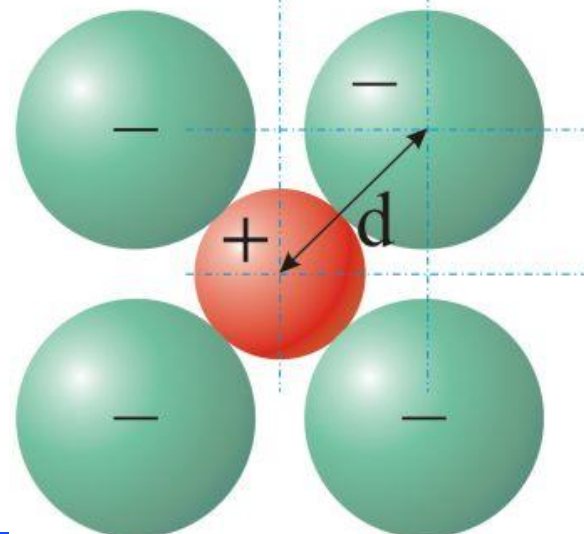
ACAP

Perovskite

$\text{CH}_3\text{NH}_3\text{PbI}_3$



CH_3NH_3

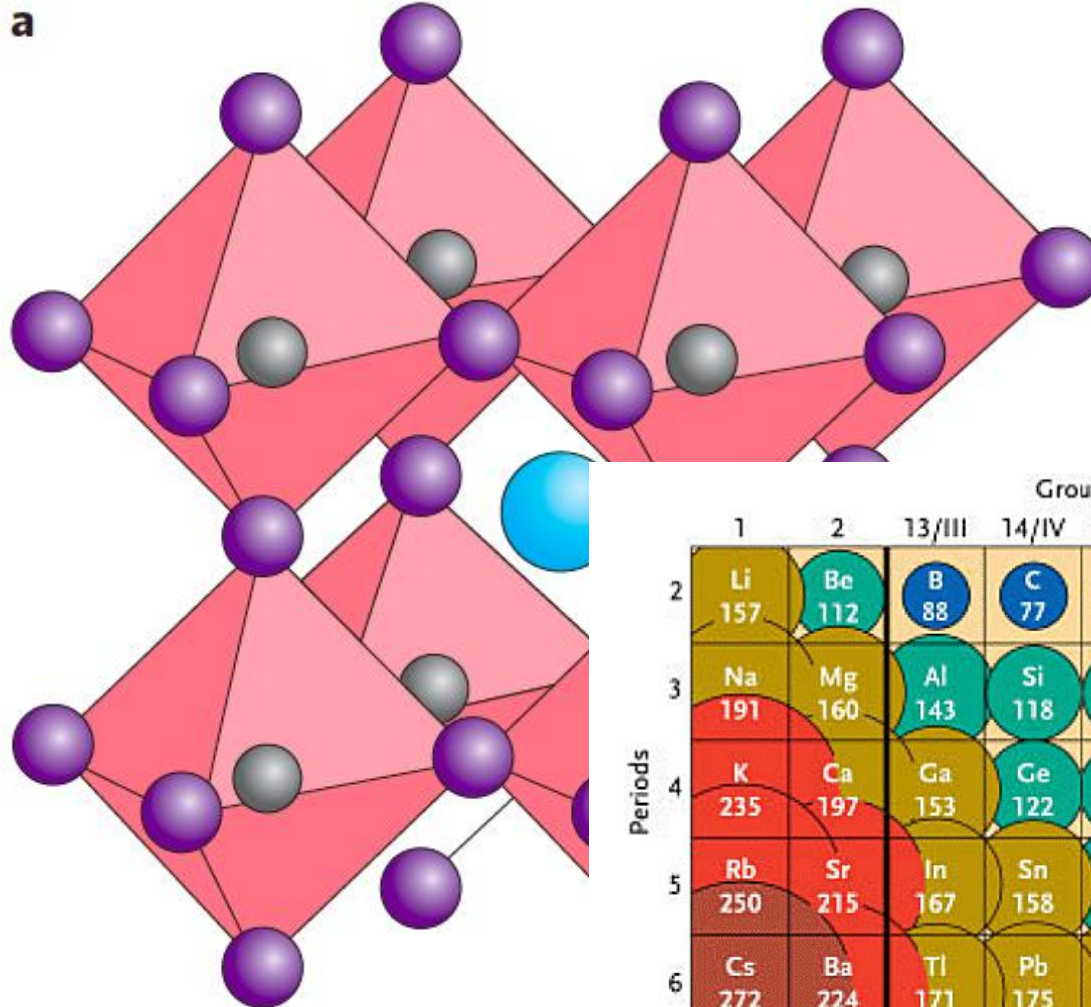


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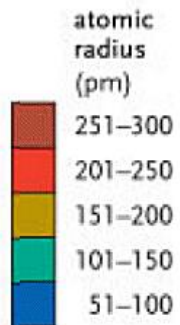
Perovskite



Groups

	1	2	13/III	14/IV	15/V	16/VI	17/VII	18/VIII
2	Li 157	Be 112	B 88	C 77	N 74	O 66	F 64	Ne
3	Na 191	Mg 160	Al 143	Si 118	P 110	S 104	Cl 99	Ar
4	K 235	Ca 197	Ga 153	Ge 122	As 121	Se 117	Br 114	Kr
5	Rb 250	Sr 215	In 167	Sn 158	Sb 141	Te 137	I 133	Xe
6	Cs 272	Ba 224	Tl 171	Pb 175	Bi 182	Po 167	At	Rn

Periods



(c) Wiley-VCH

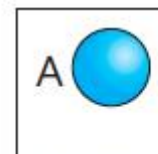
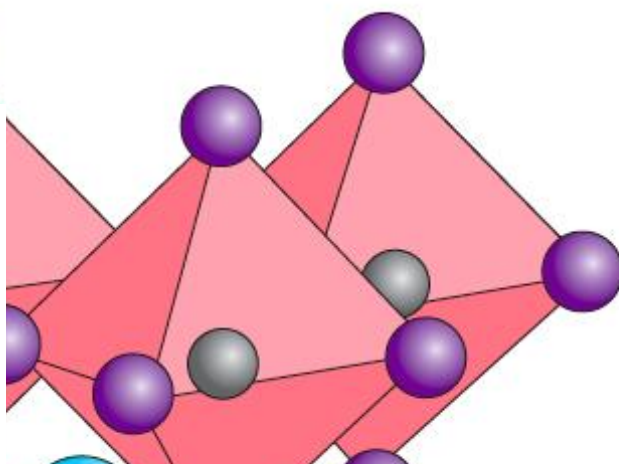
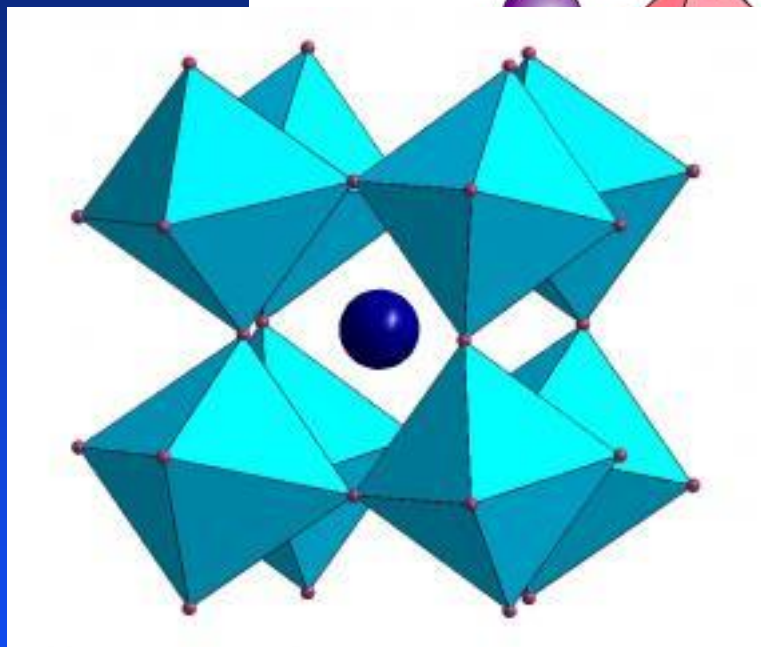
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Perovskite

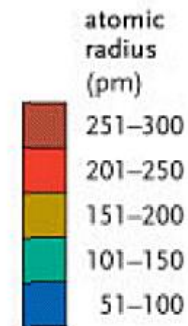
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Periods

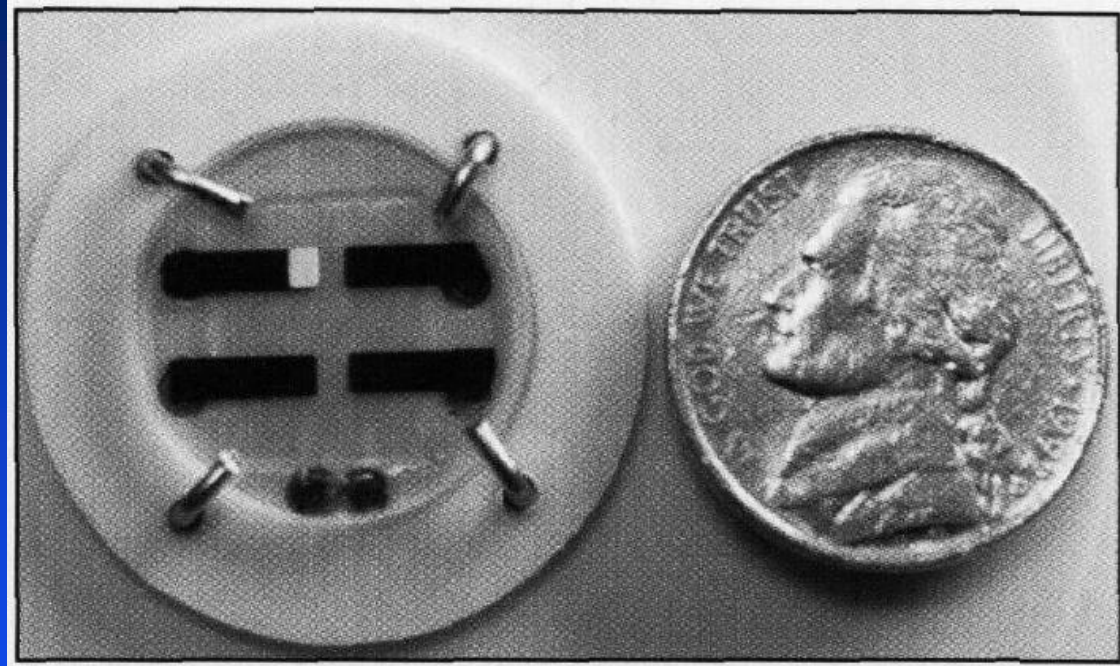


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History



David Mitzi then at IBM

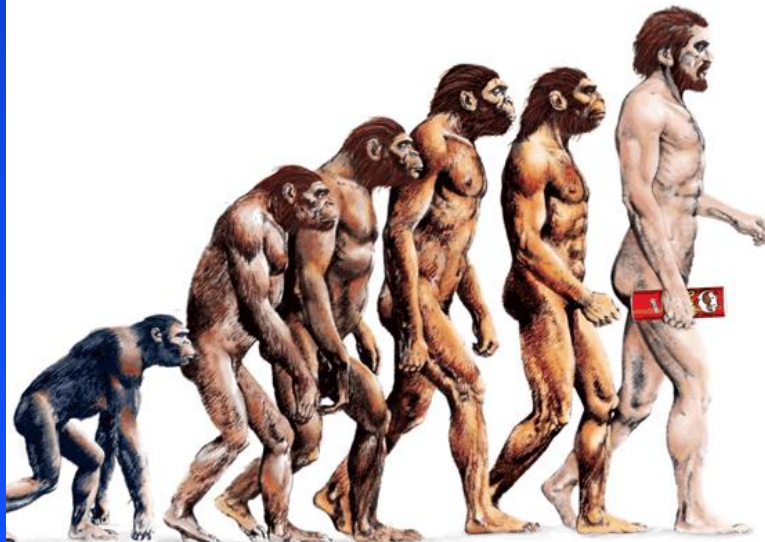
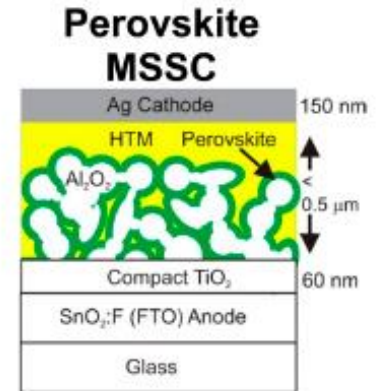
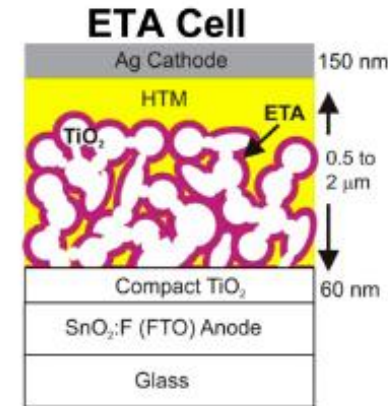
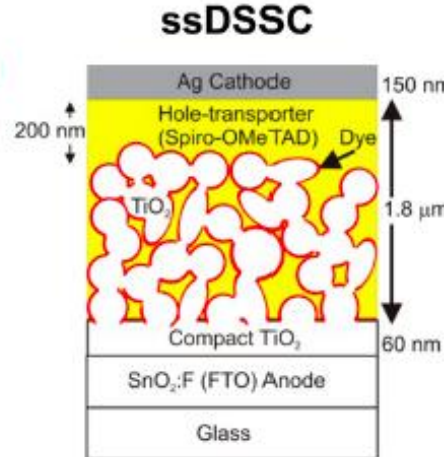
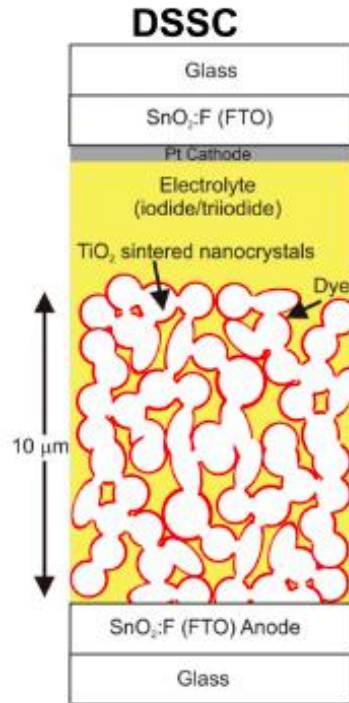
From 1995 ~ 2002, investigated use in LEDs and FETs

Pb toxicity issue for PV as well as stability – even worse for Sn compounds

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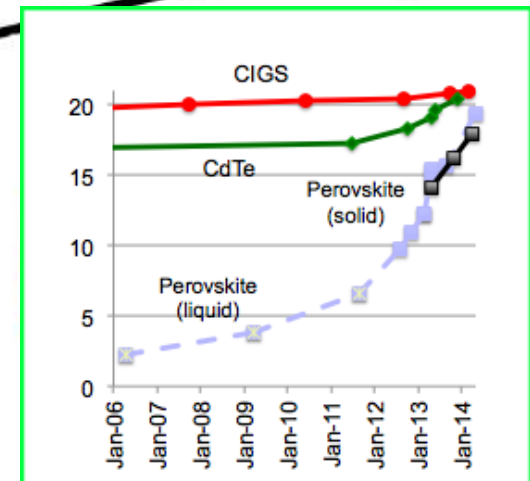
Perovskite Evolution

Snaith, H.J. Phys. Chem. Lett. 2013, 4, 3623



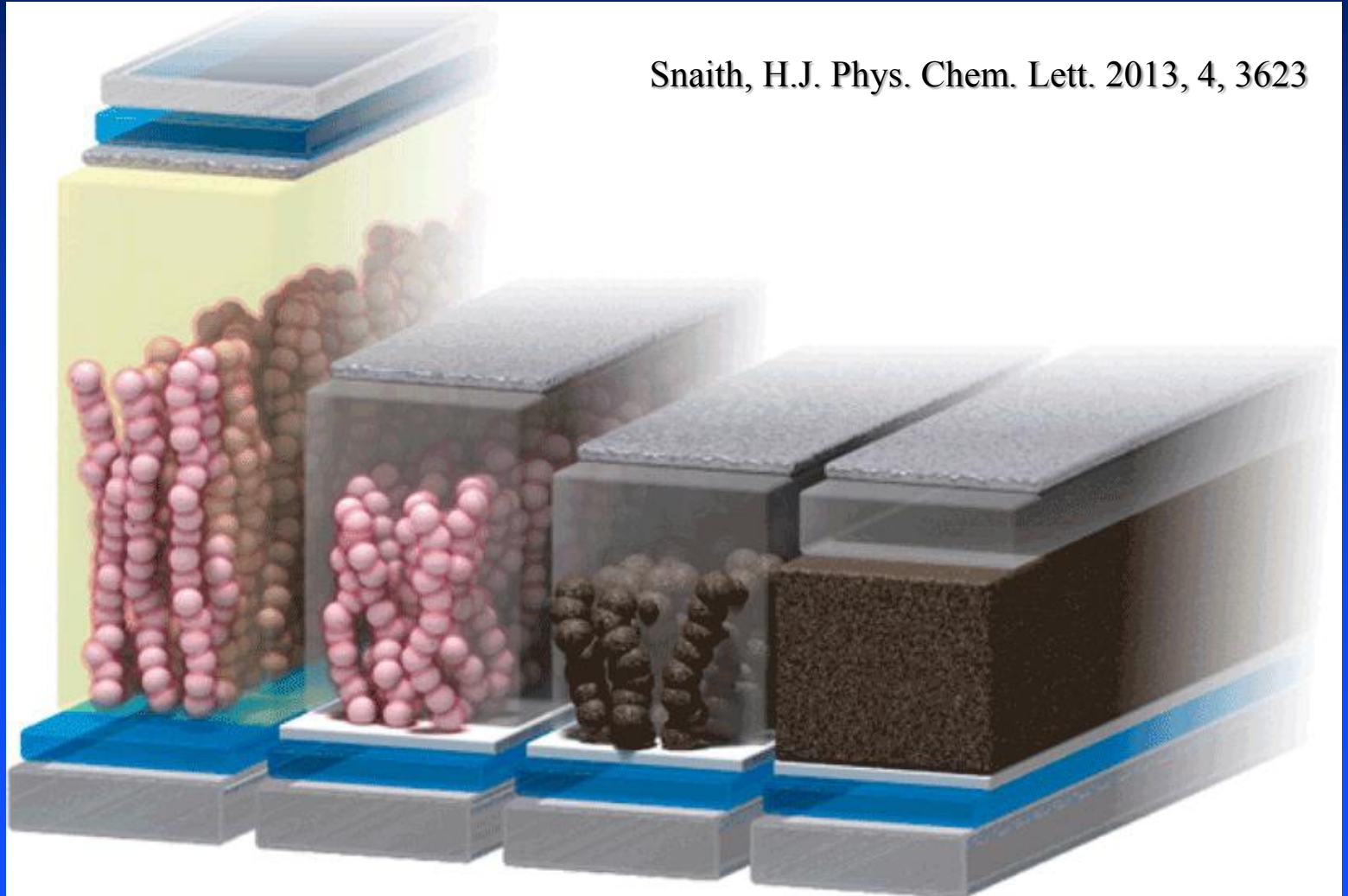
Future direction

"p-i-n" thin-film perovskite

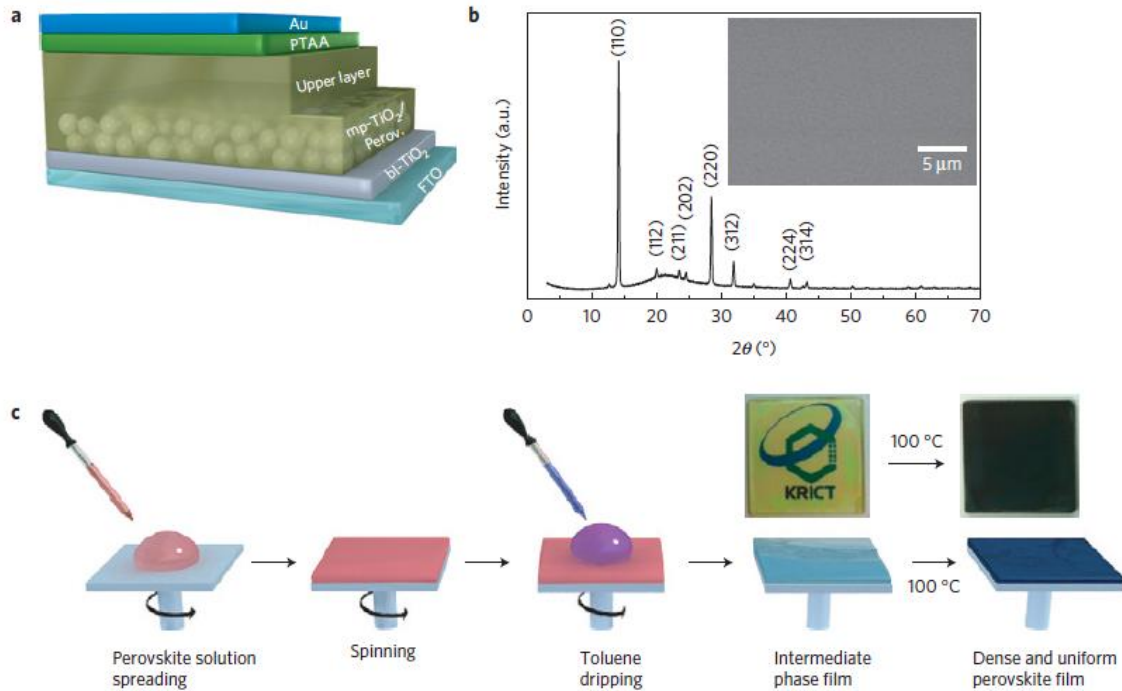


Perovskite Evolution

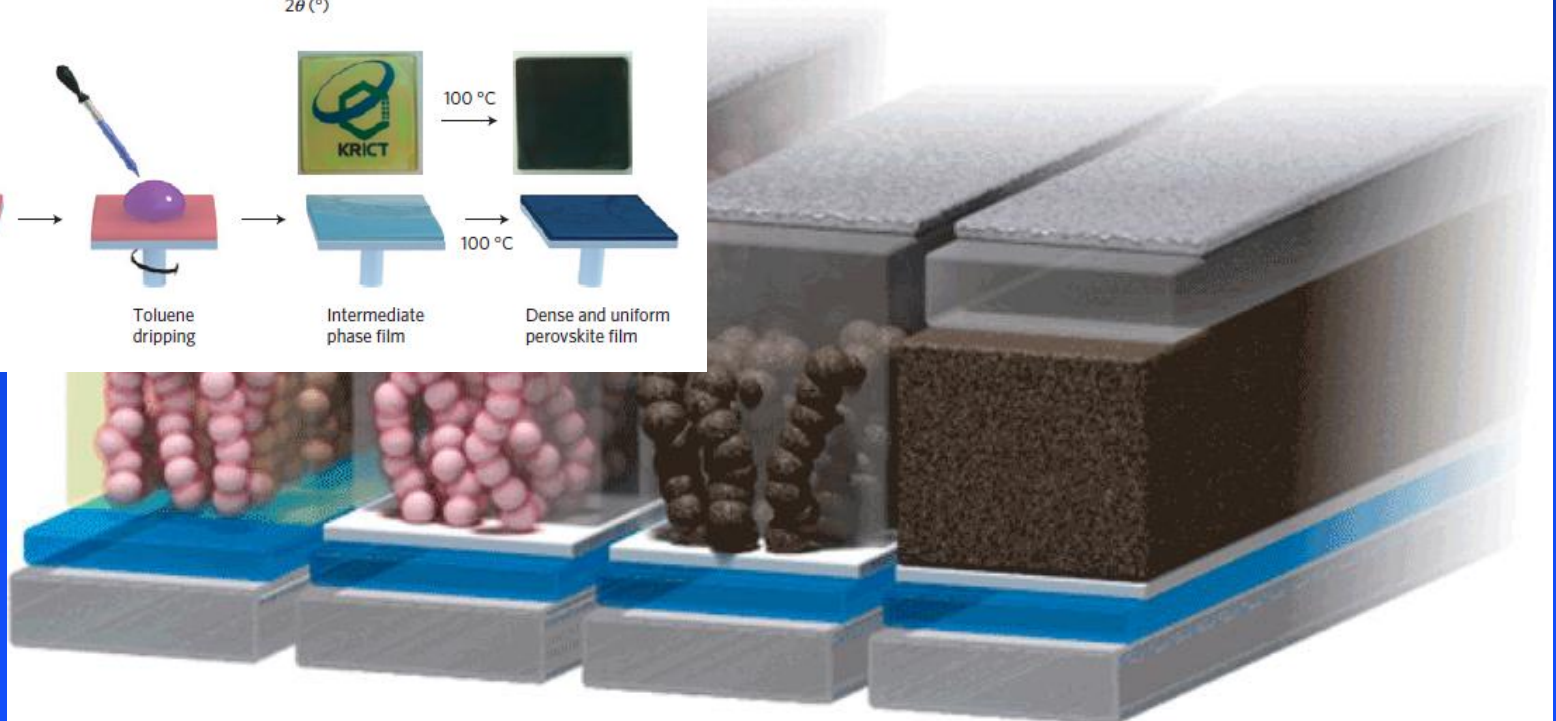
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Liquids



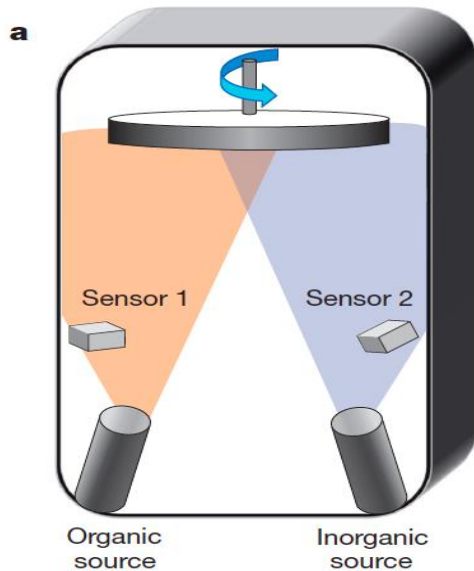
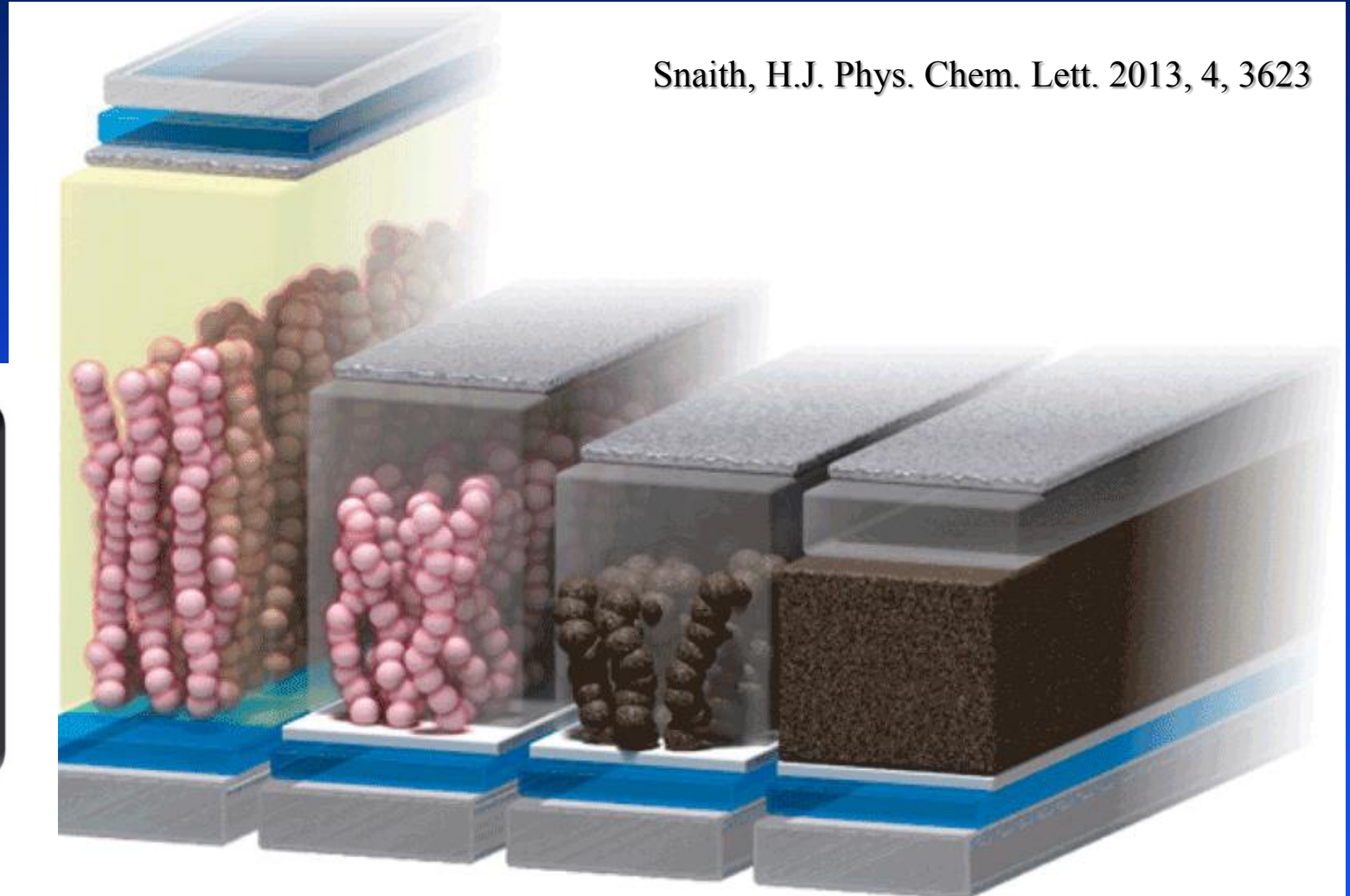
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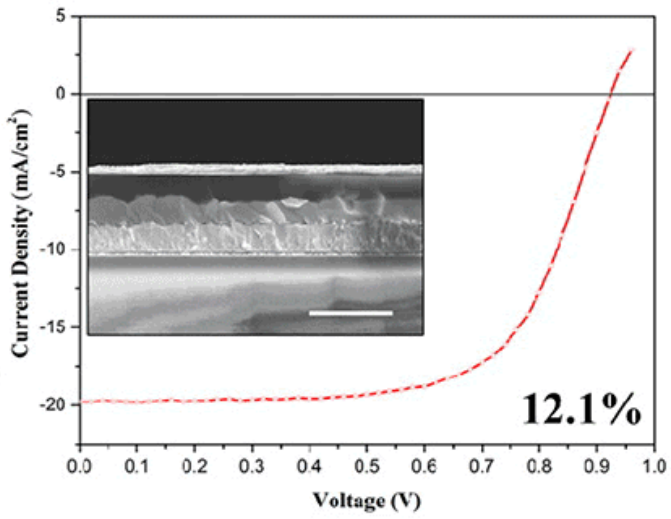
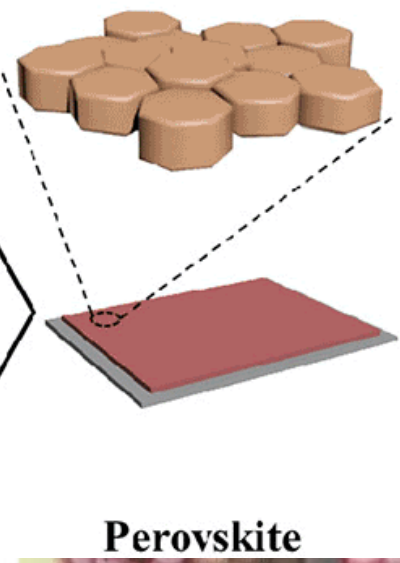
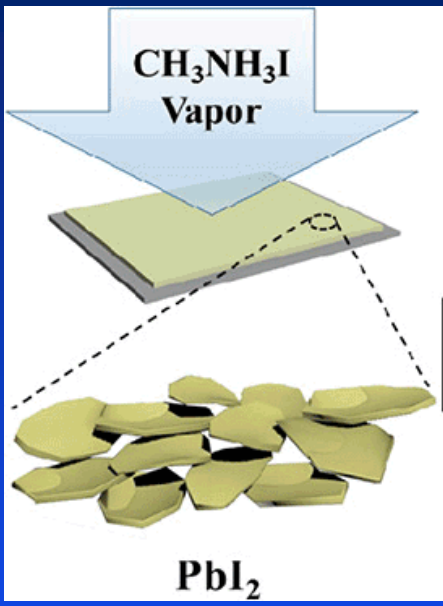
Evaporation

Snaith, H.J. Phys. Chem. Lett. 2013, 4, 3623



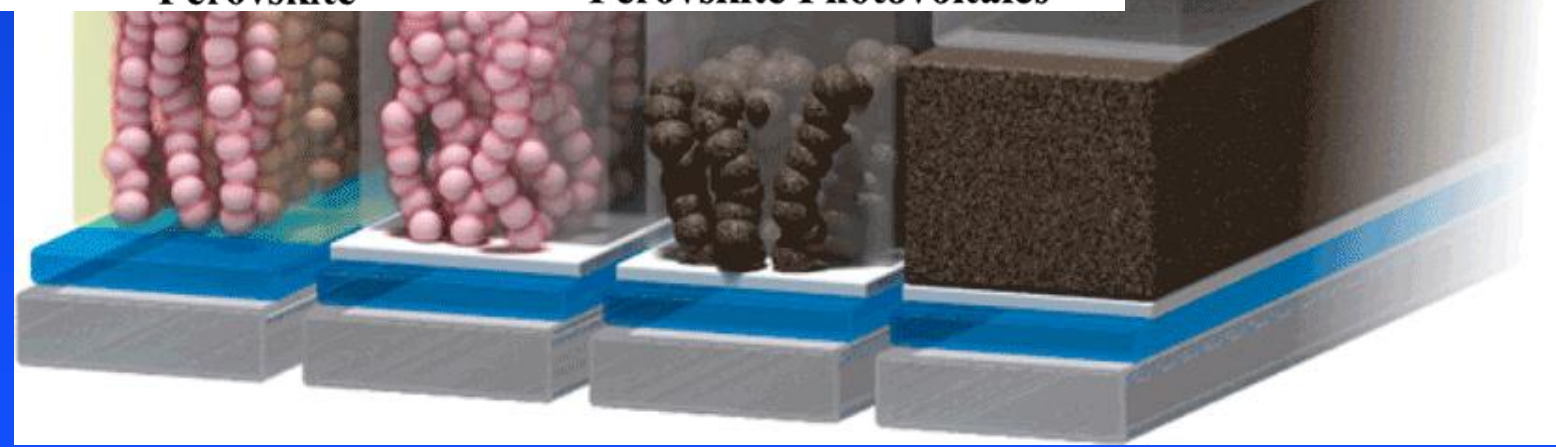


Vapour phase



em. Lett. 2013, 4, 3623

Perovskite Photovoltaics

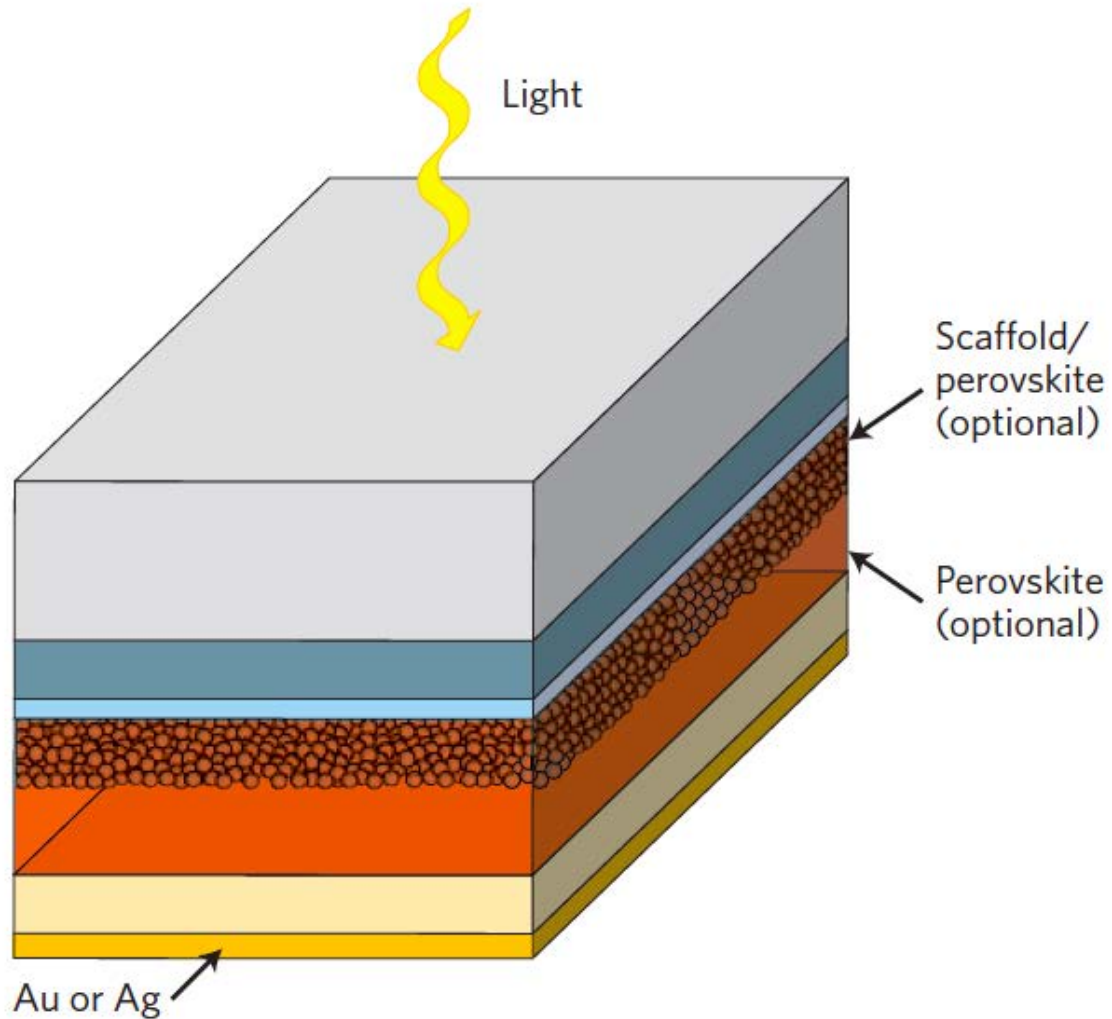




ACAP

Device structure

- Glass
- FTO
- Compact TiO₂
- Perovskite
- HTM
- Au

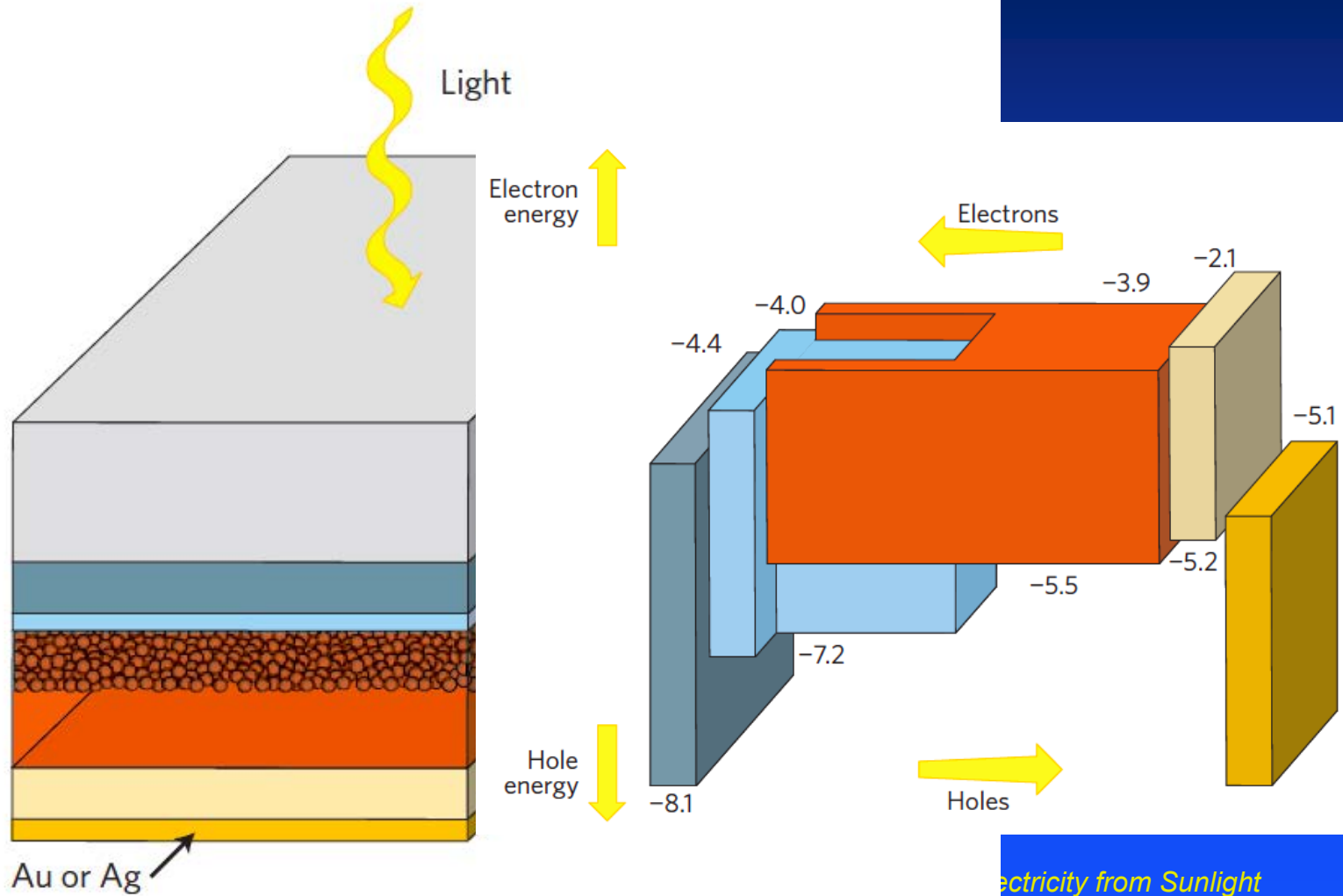




ACAP

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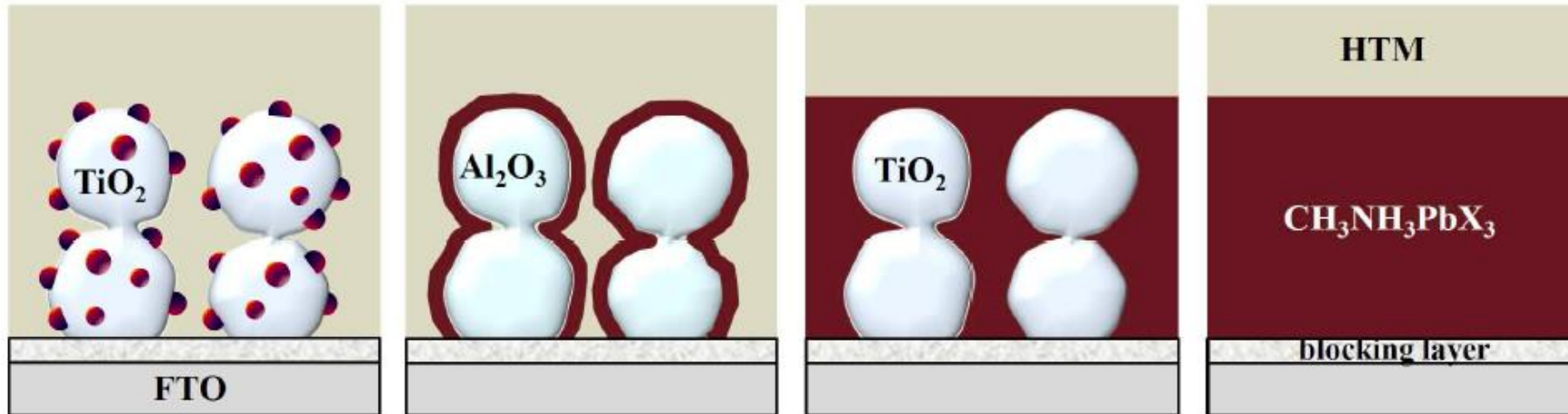


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Electricity from Sunlight



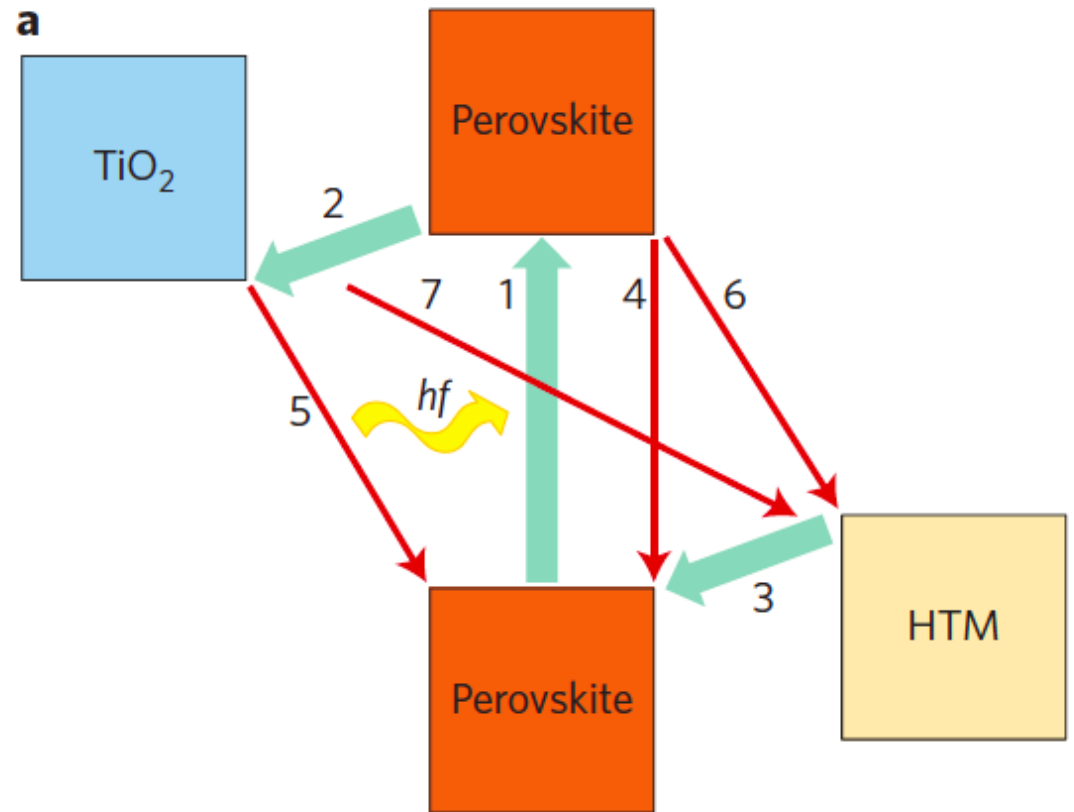
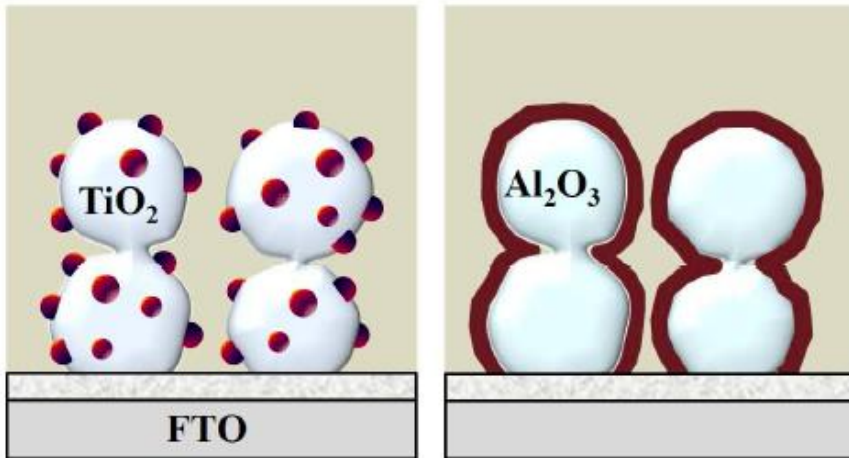
Operating principles



HS Kim, SH Im, and N-G Park J. Phys. Chem. C, DOI: 10.1021/jp409025w

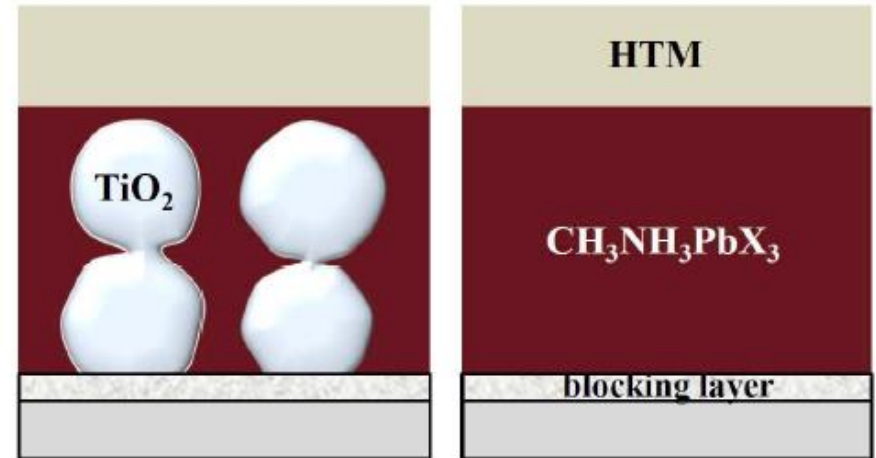
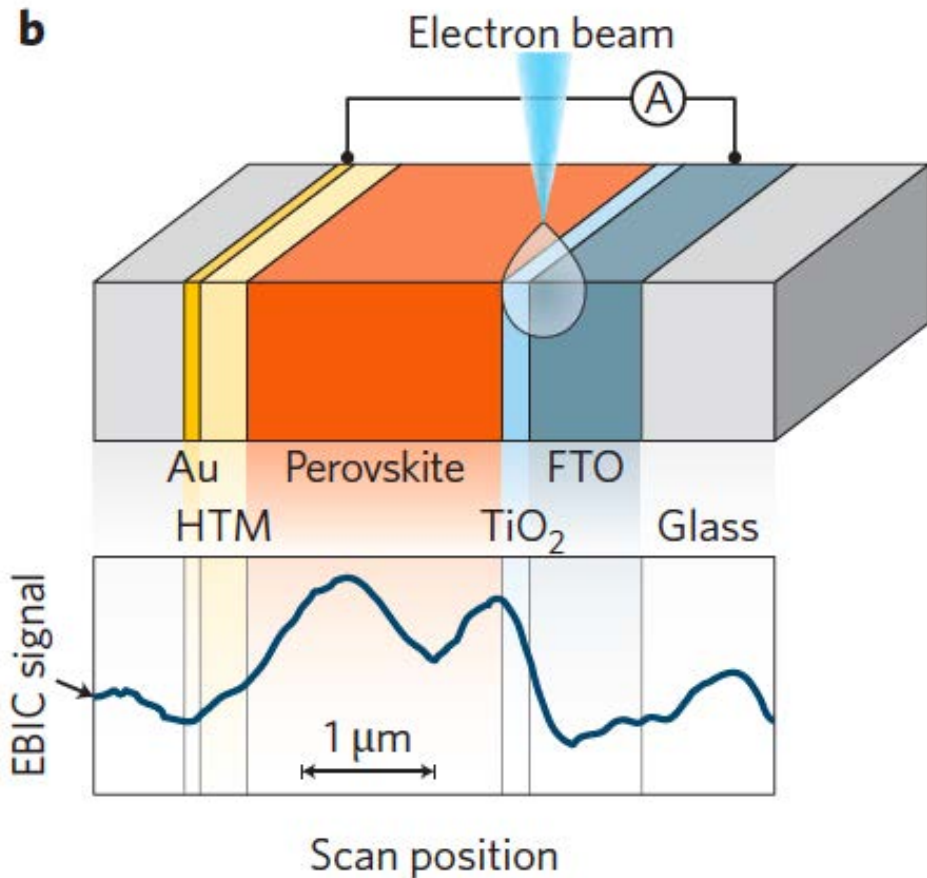


Operating principles



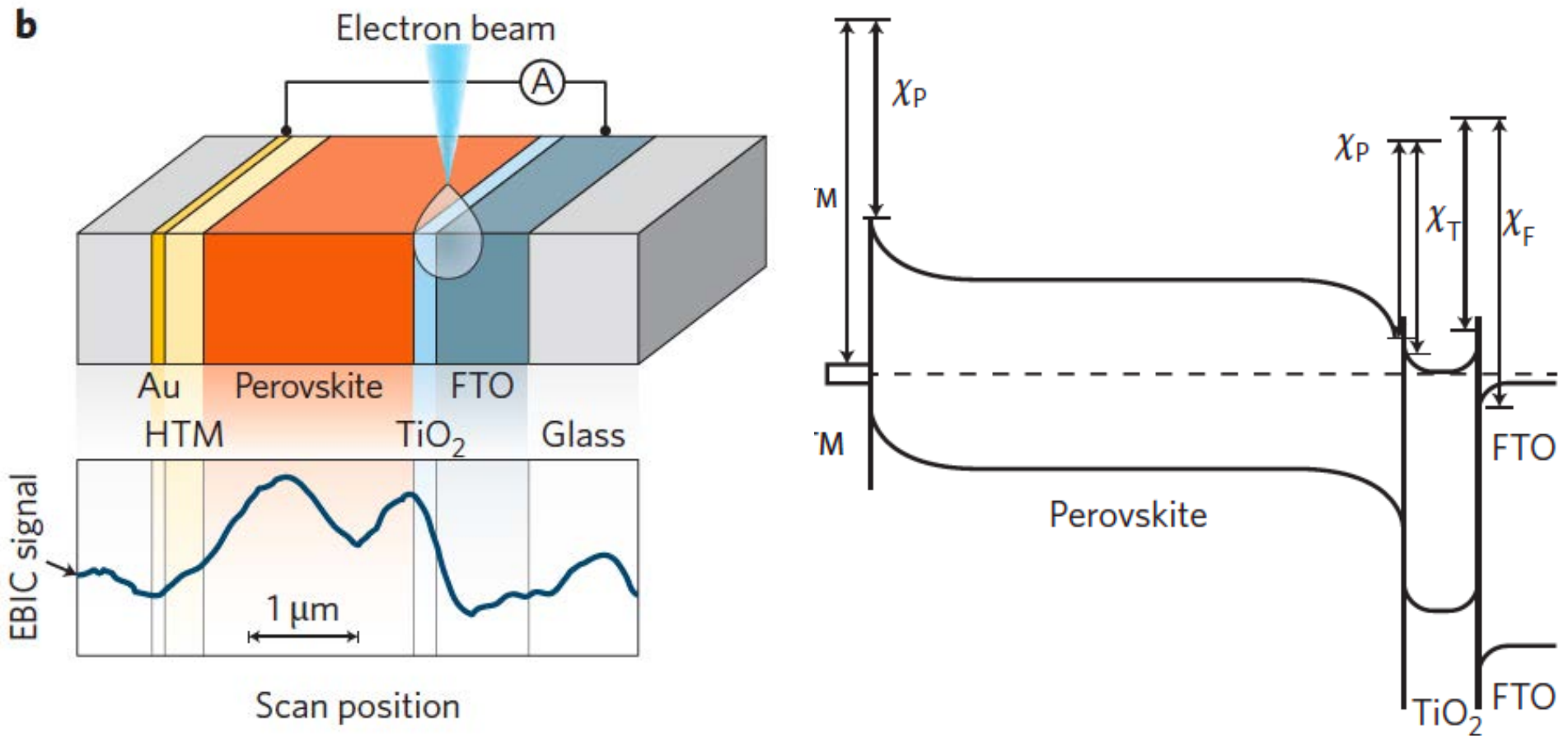


Operating principles



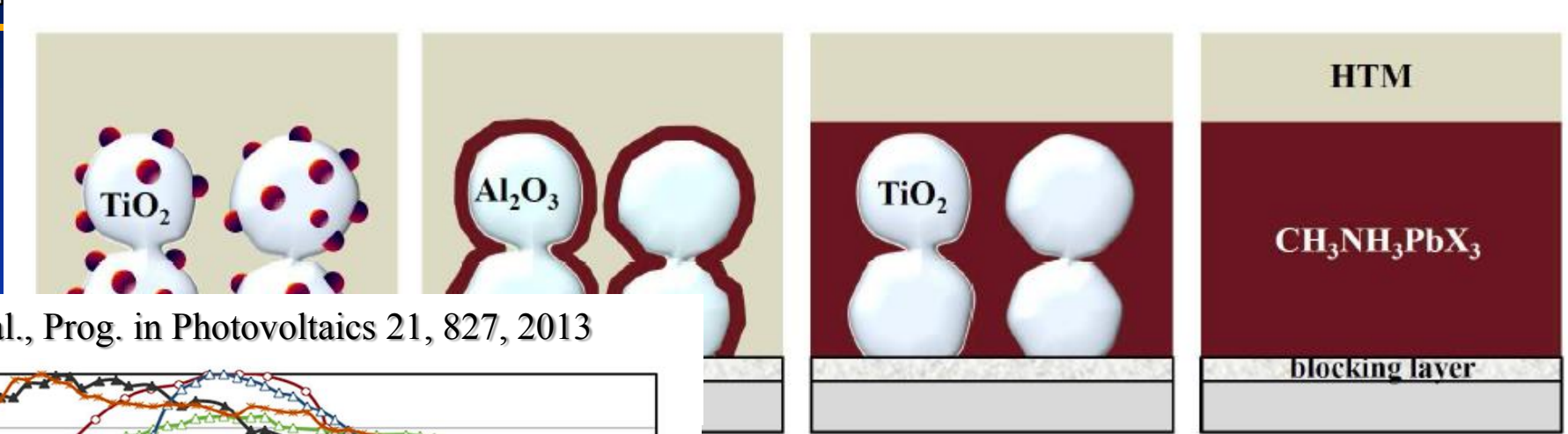


Operating principles

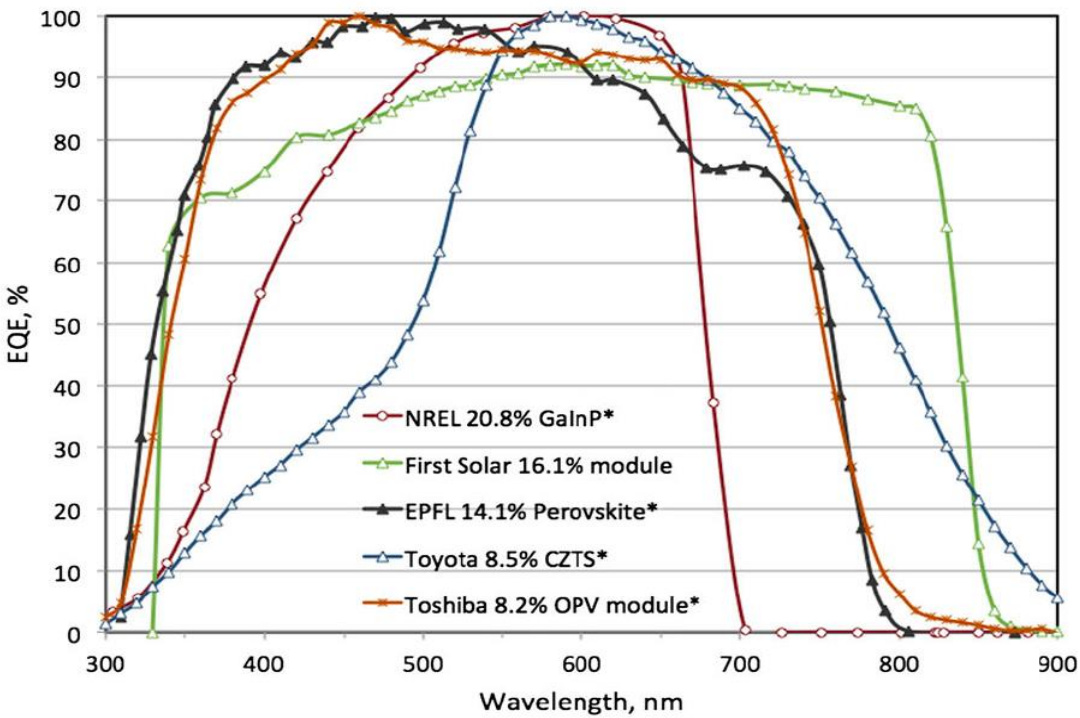




High Voc – low recombination

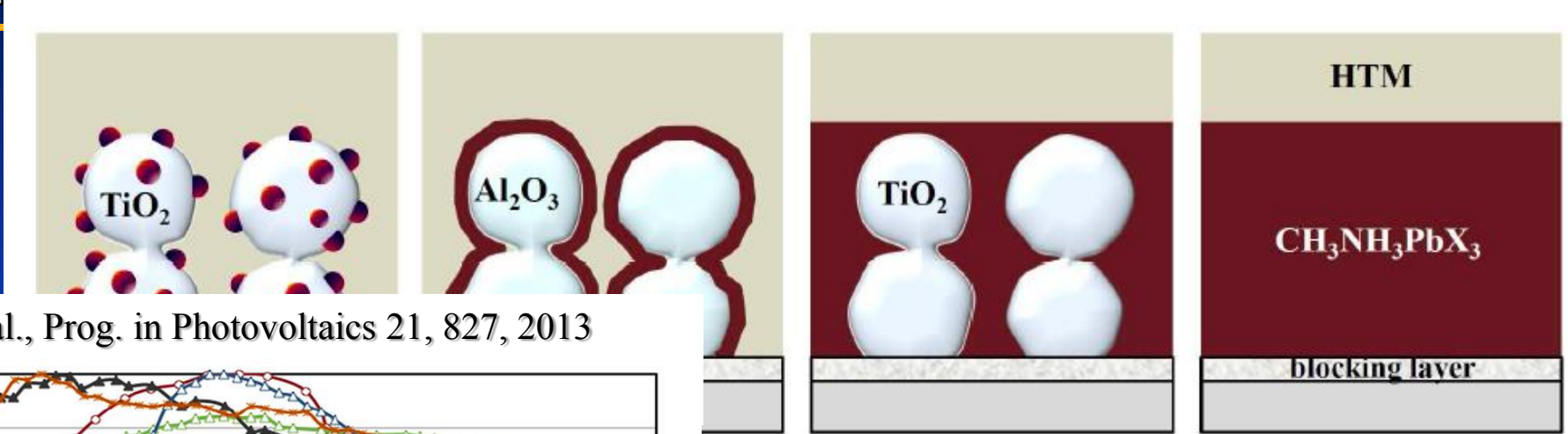


MA Green et al., Prog. in Photovoltaics 21, 827, 2013

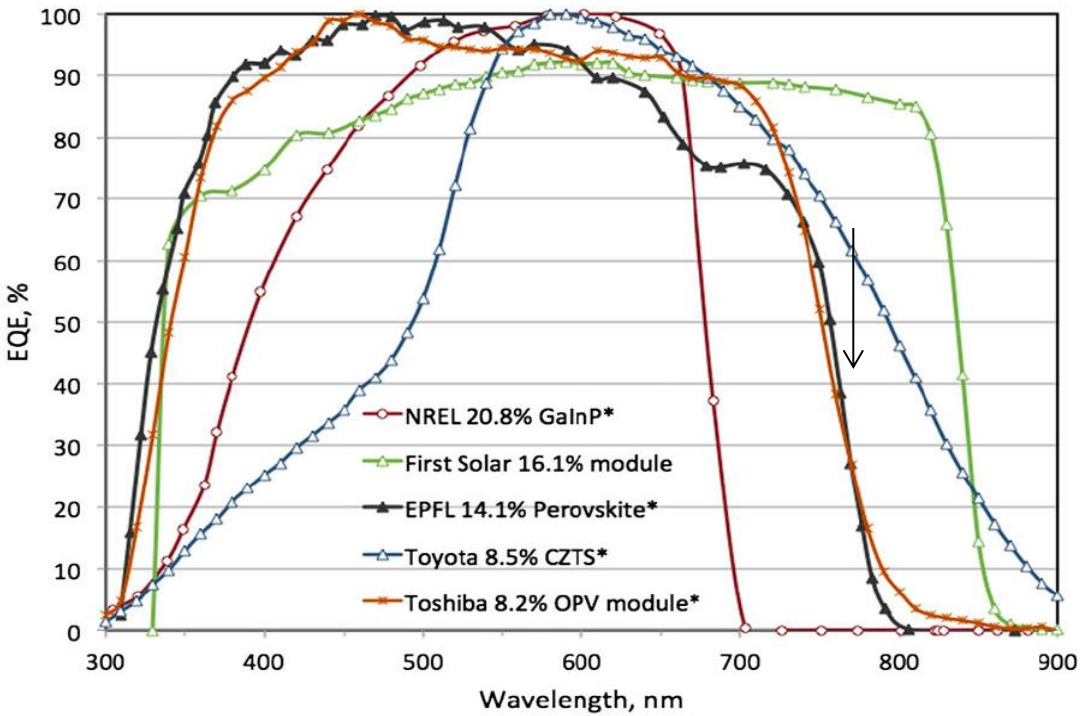




High Voc – low recombination

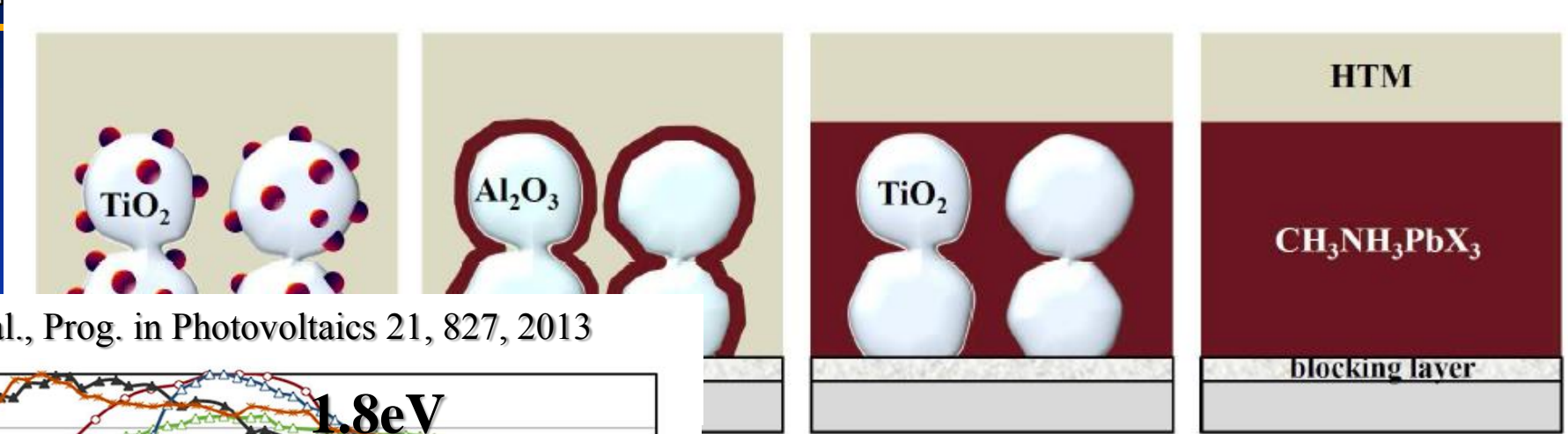


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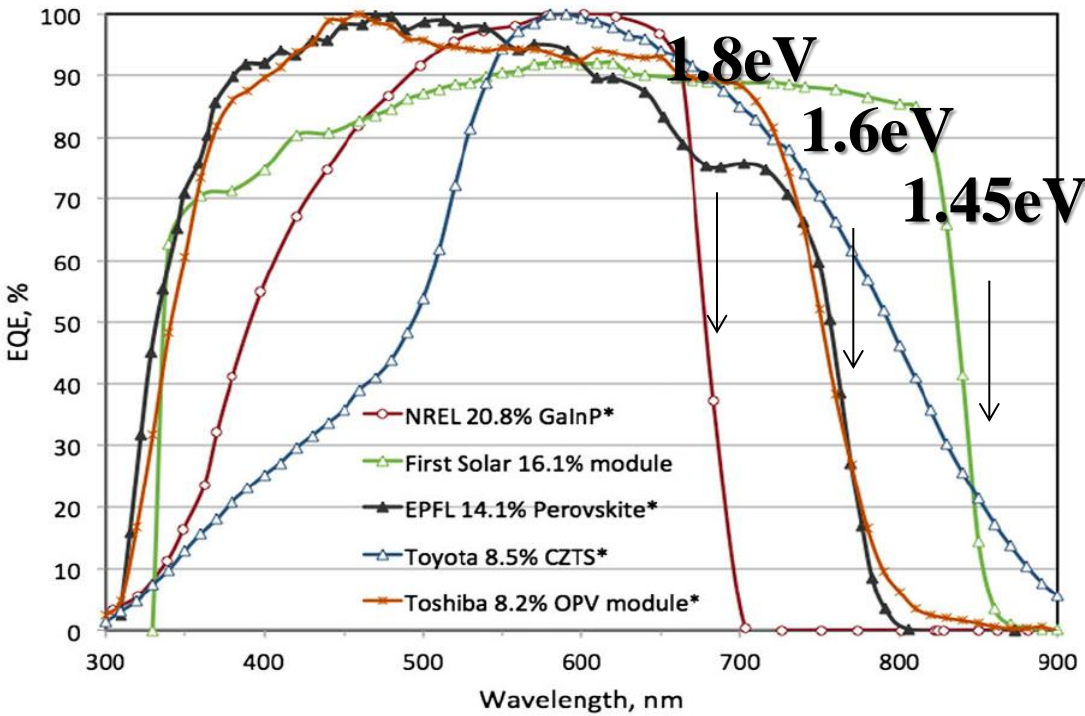




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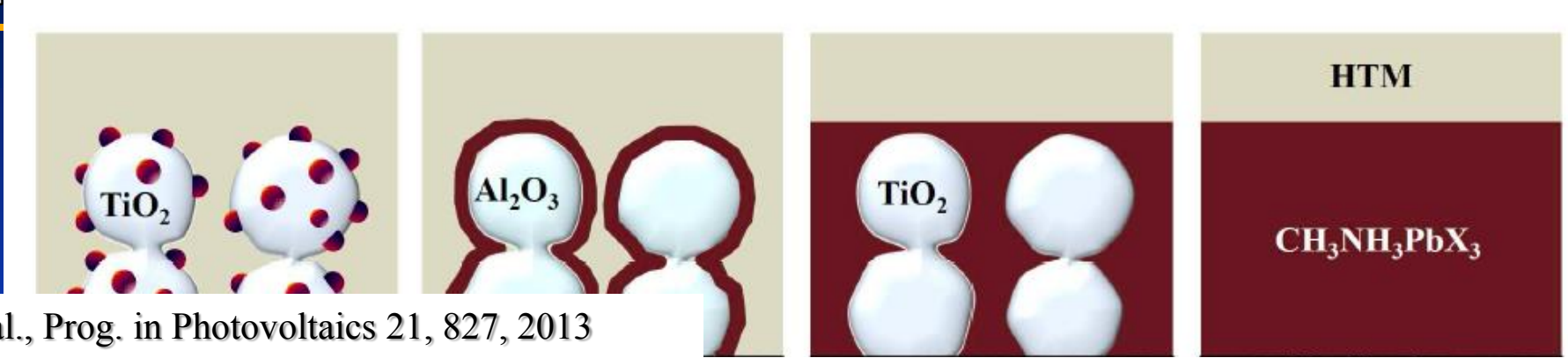


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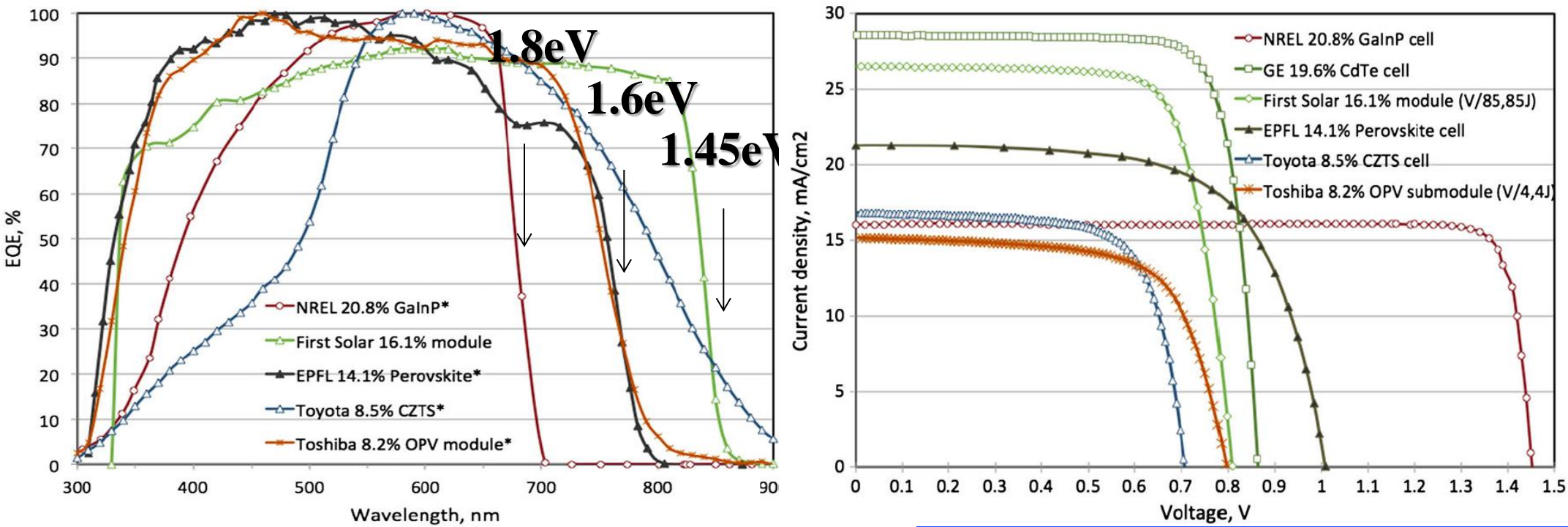




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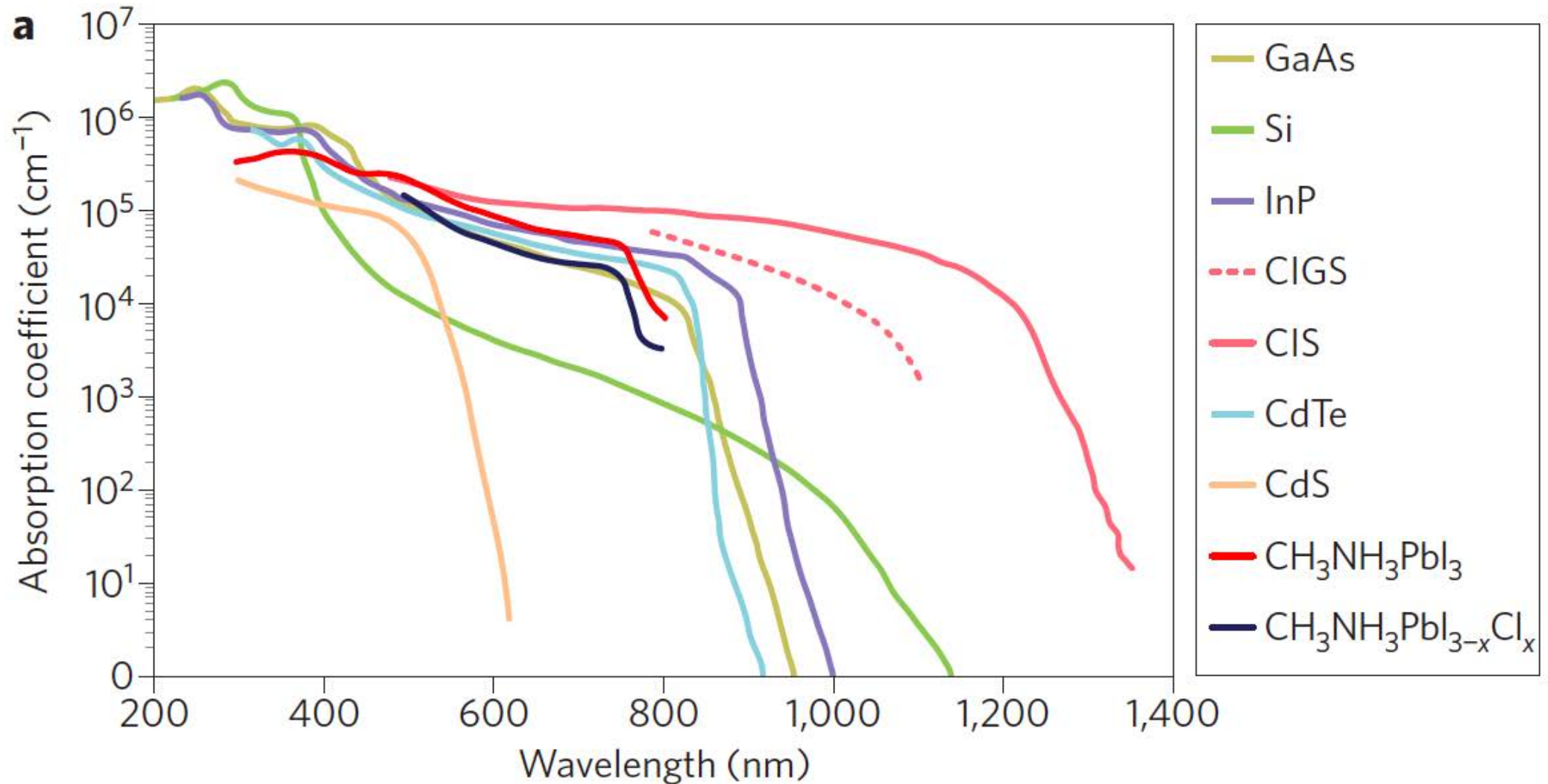
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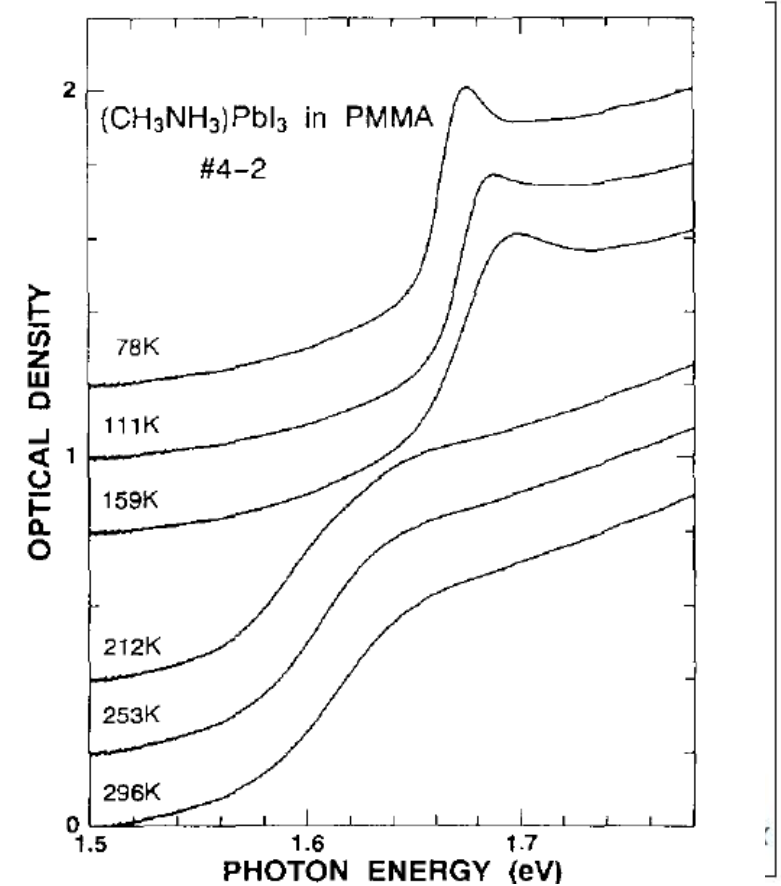
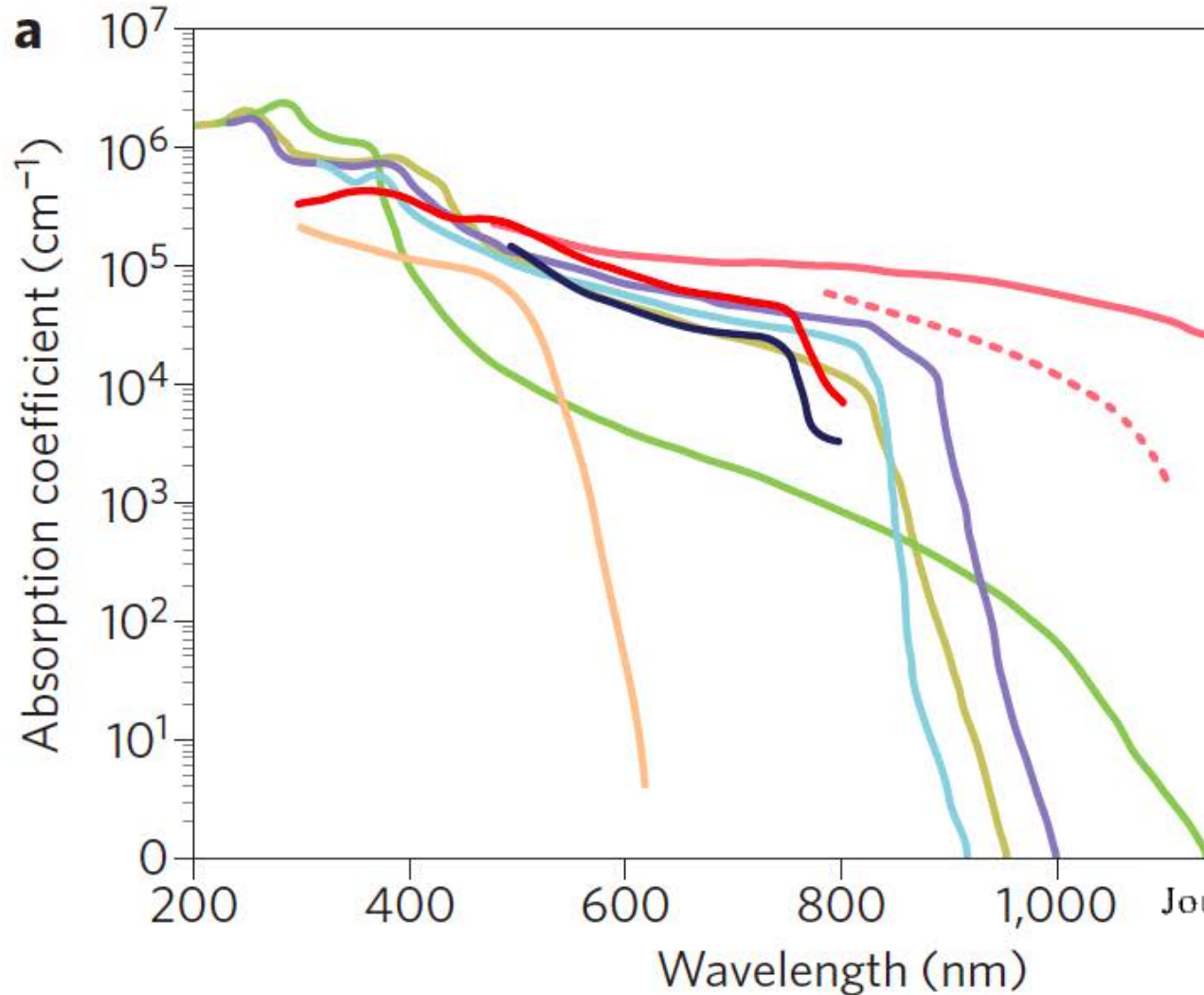
Strong absorption





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Strong absorption





Strong absorption

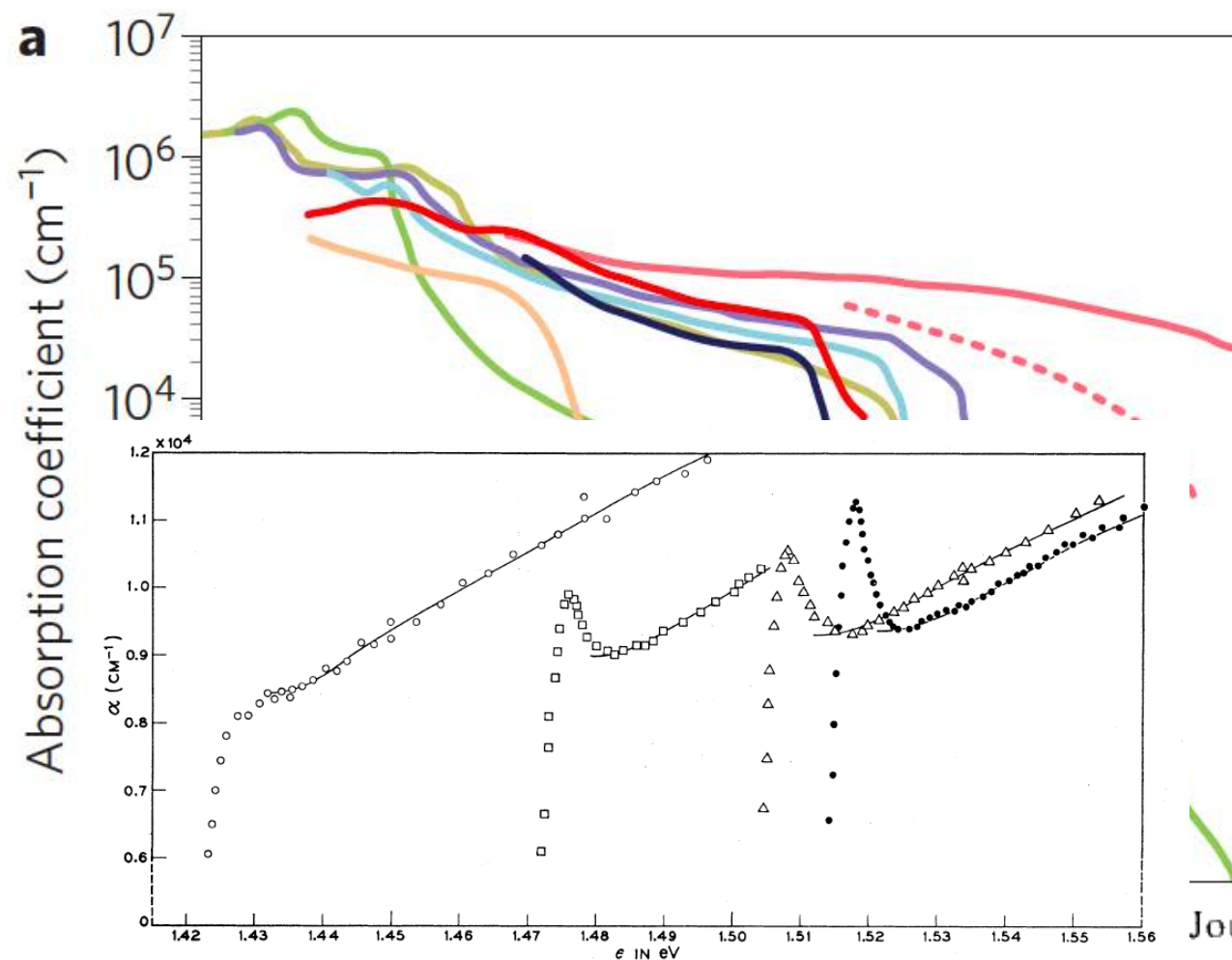
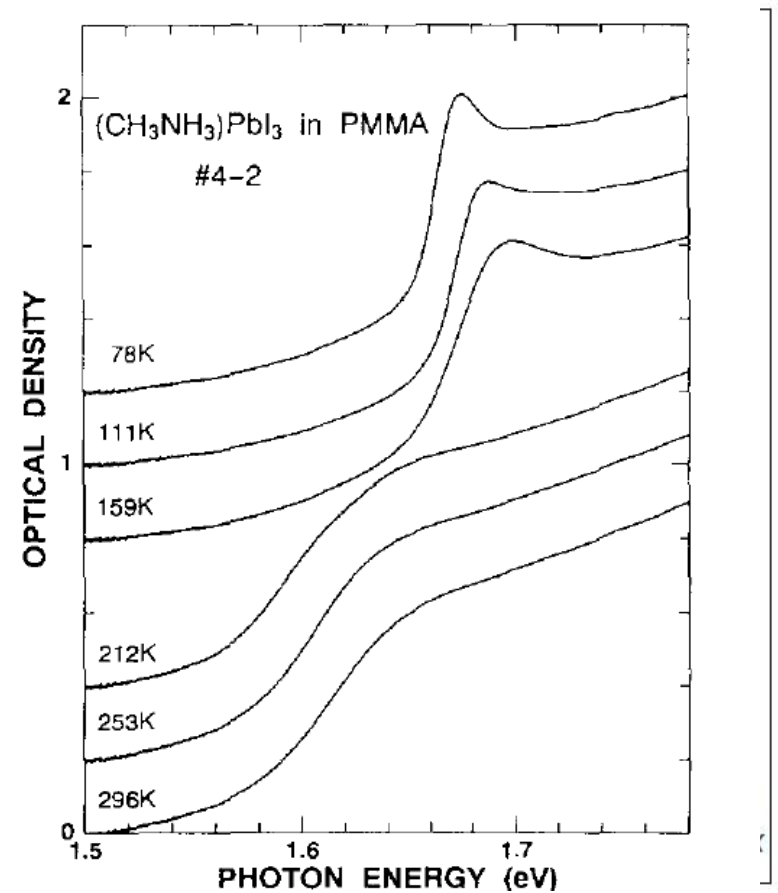


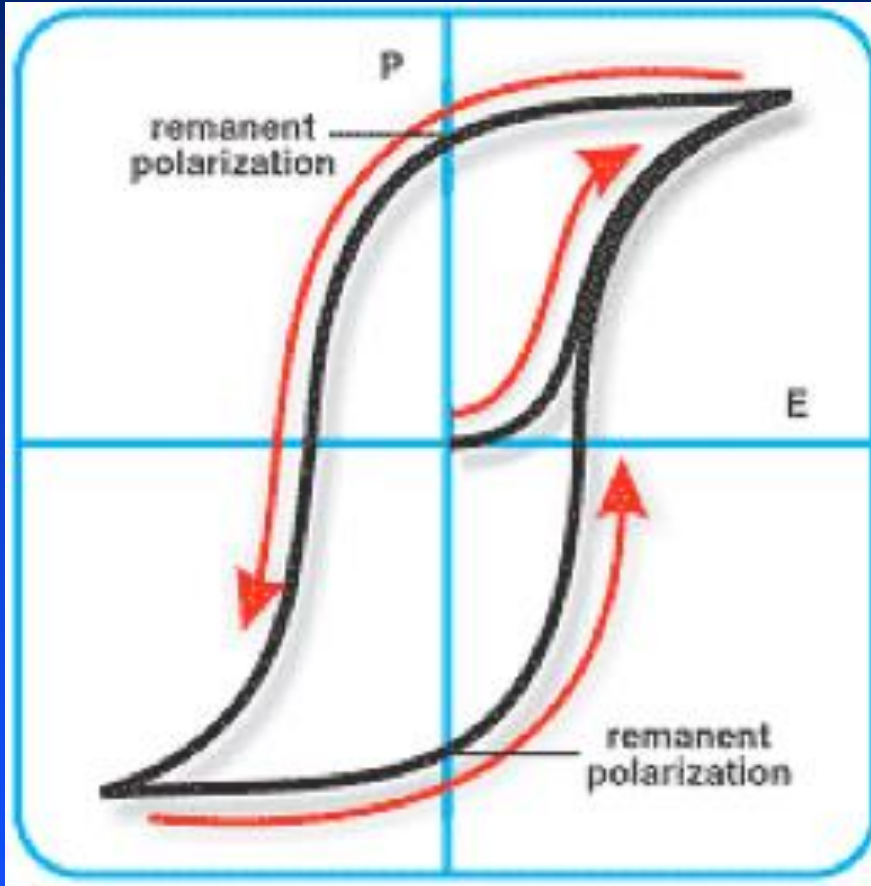
FIG 3 Exciton absorption in GaAs; ○ 294°K, □ 186°K, △ 190°K, ● 21°K.





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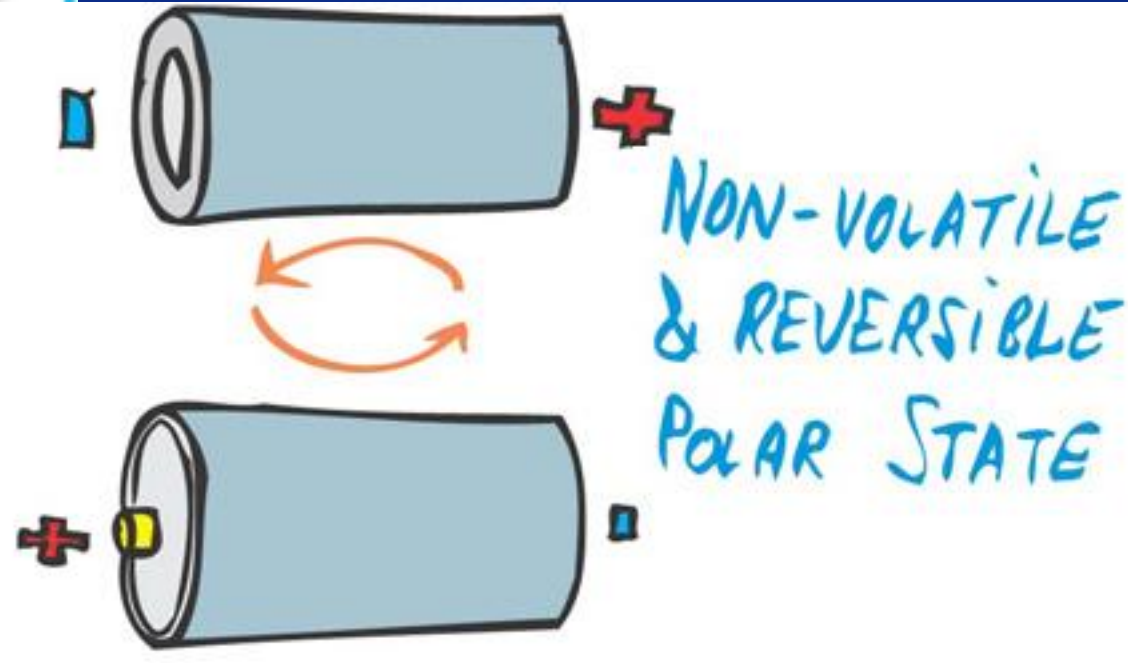
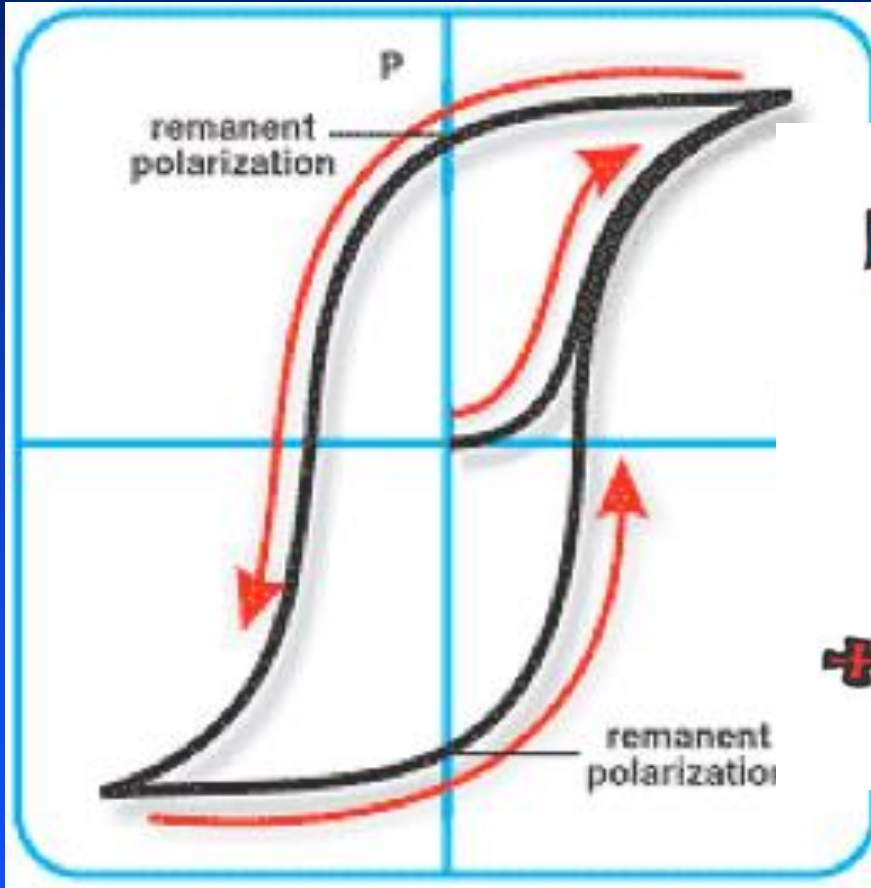
Ferroelectric?





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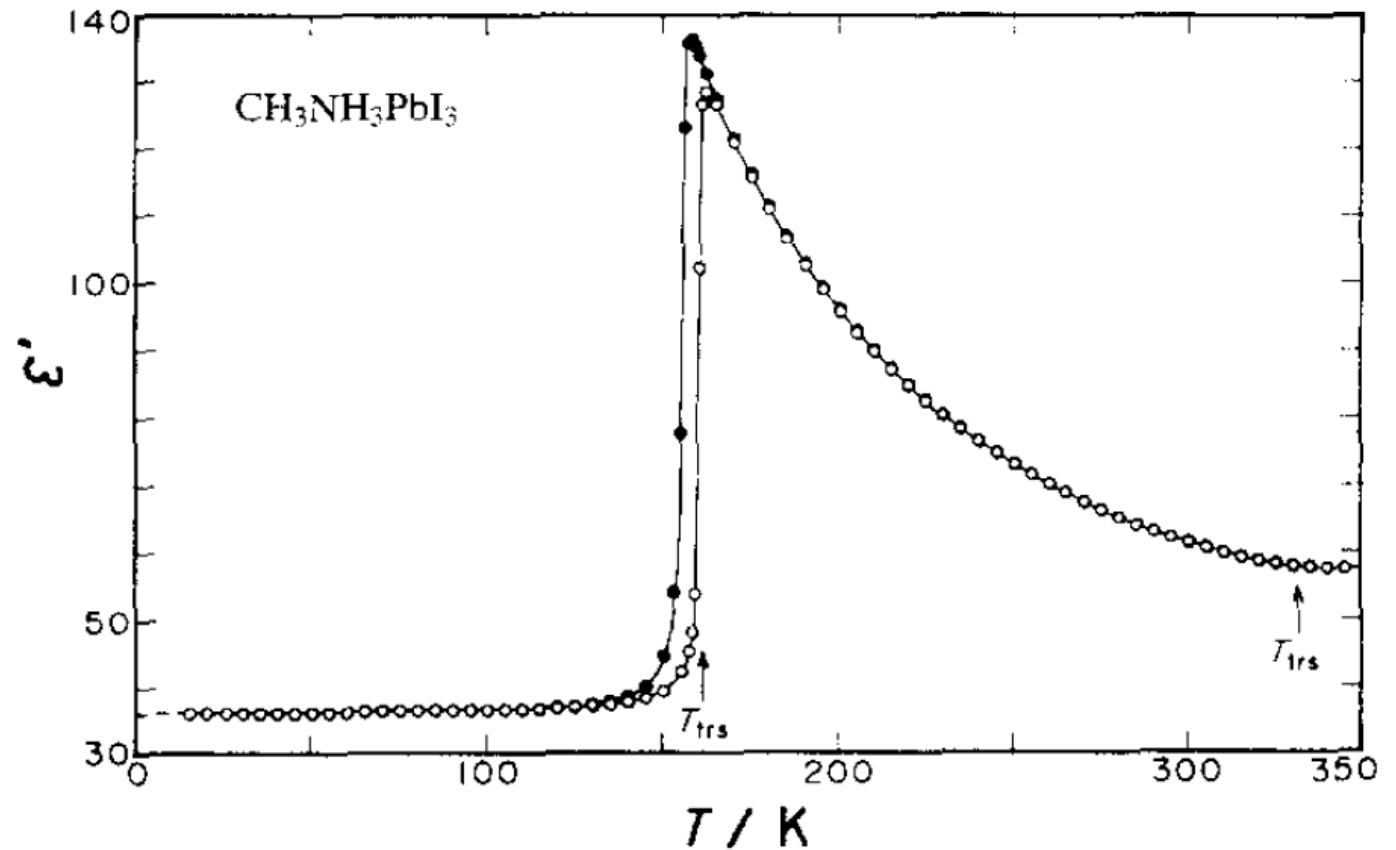
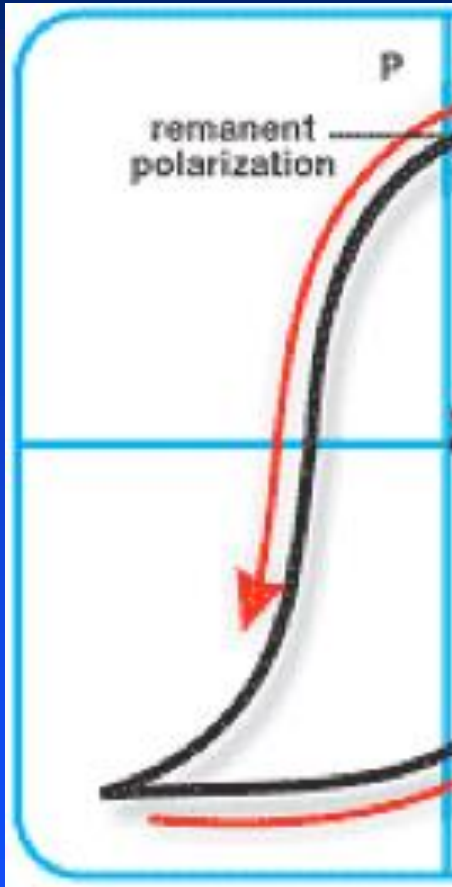
Ferroelectric?





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Ferroelectric?



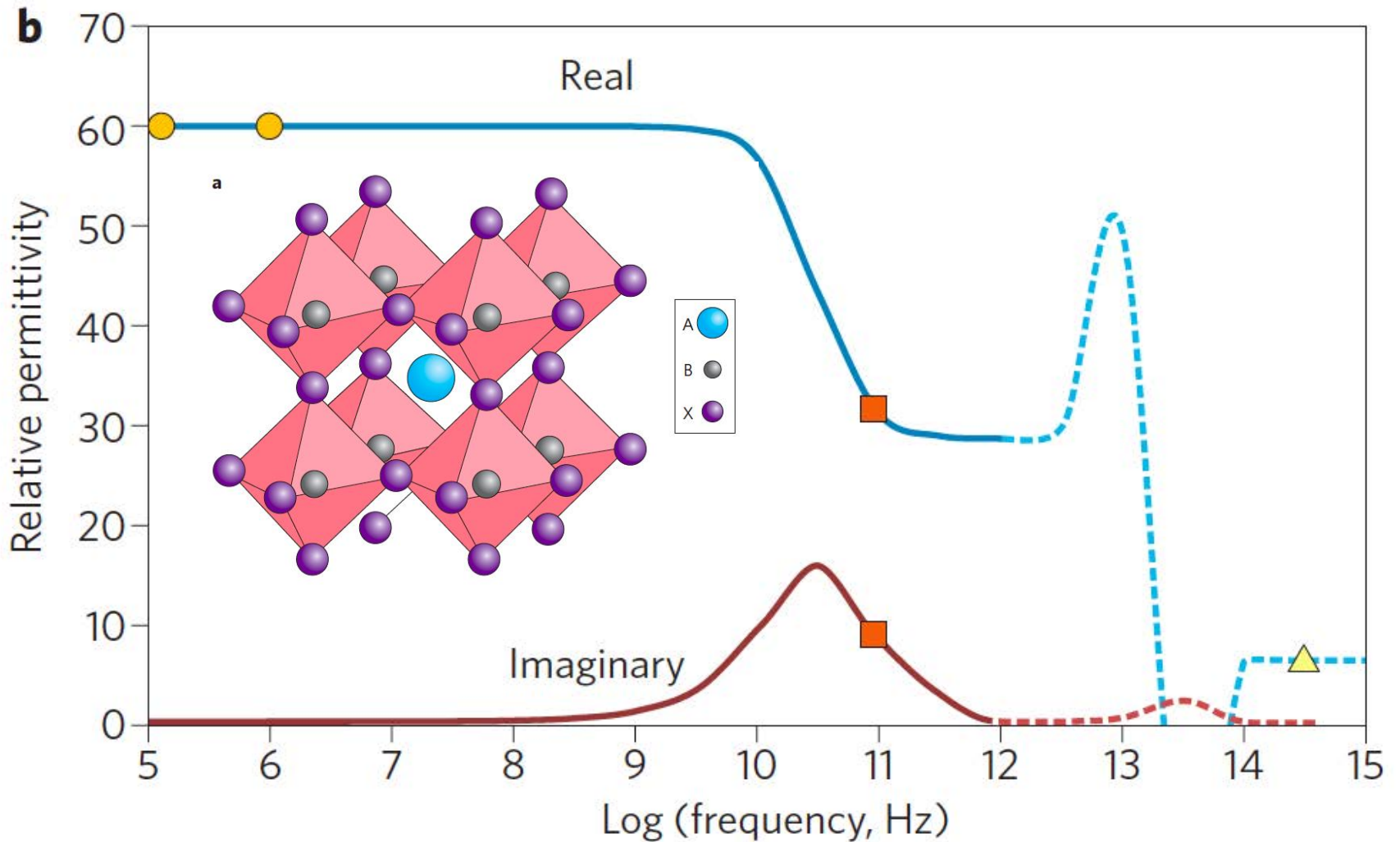
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J. Phys. Chem. Solids Vol. 53, No. 7, pp. 935–939, 1992



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Ferroelectric?

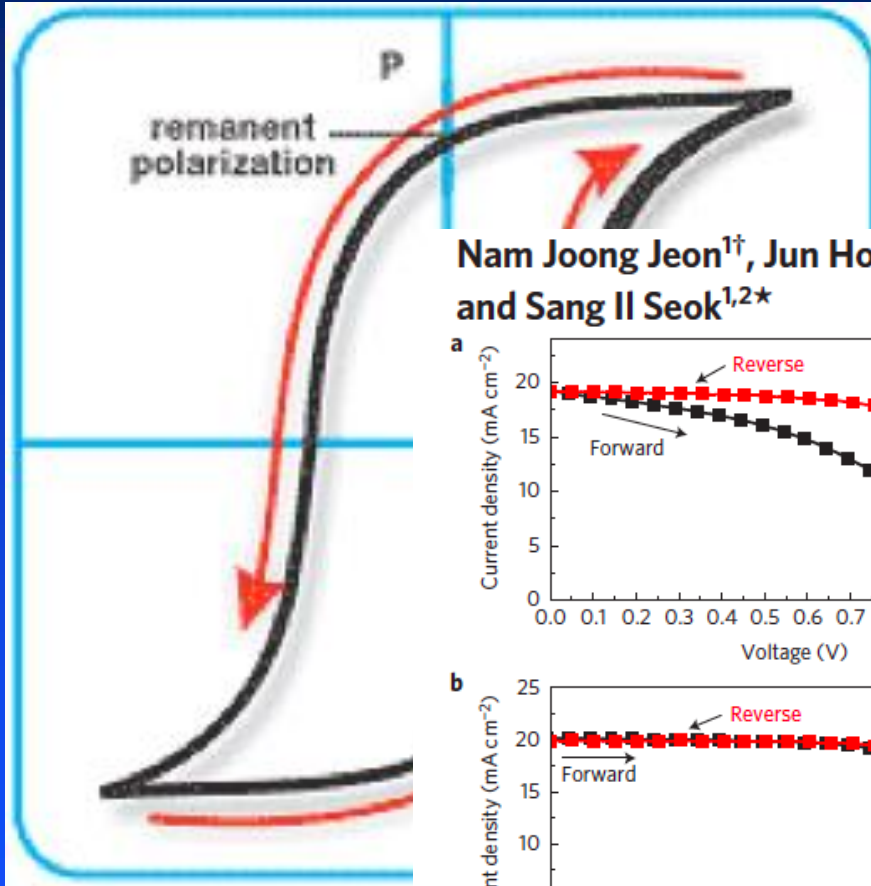


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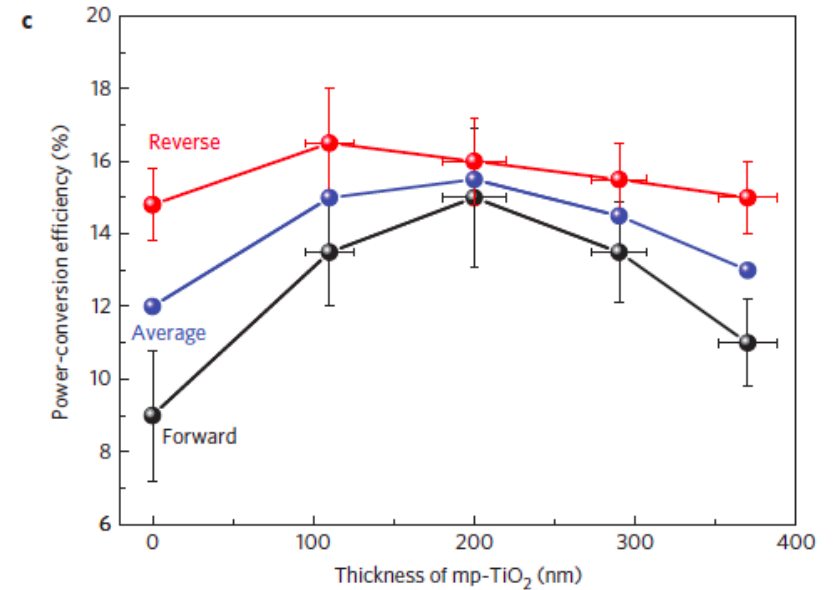
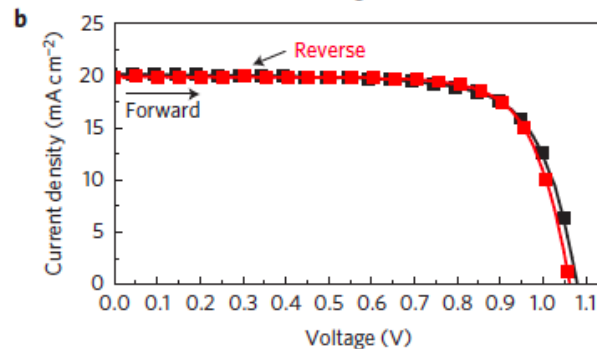
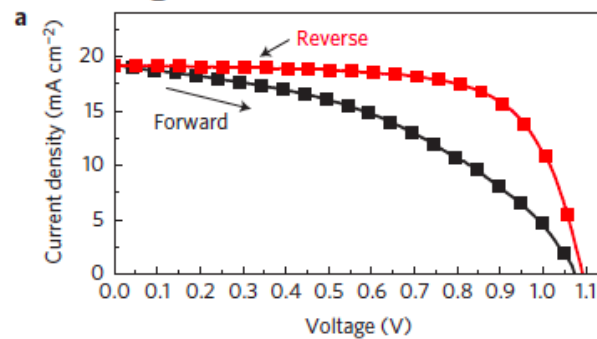


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Ferroelectric?

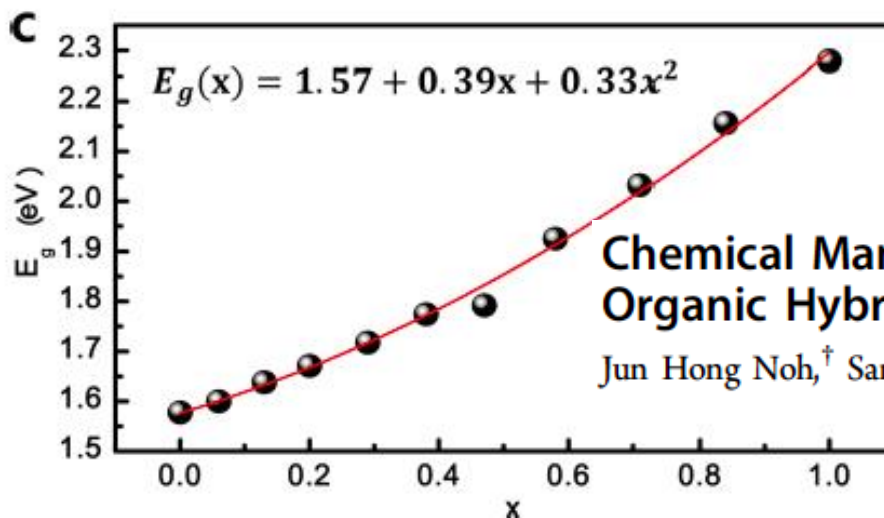
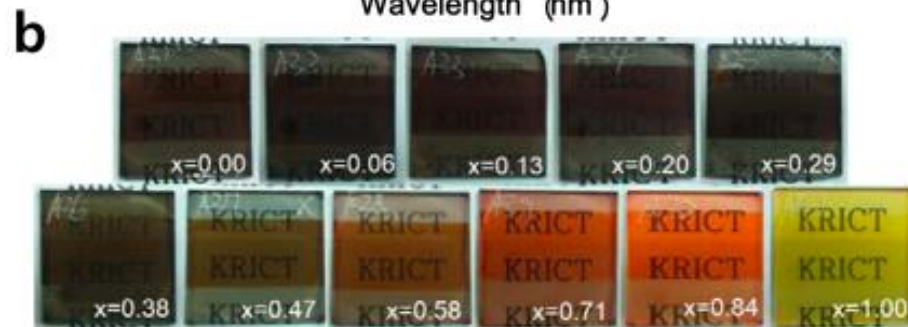
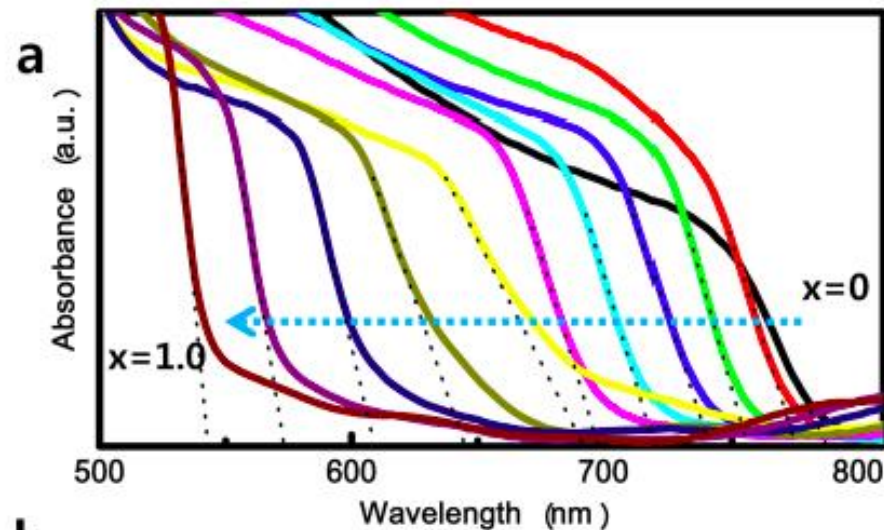
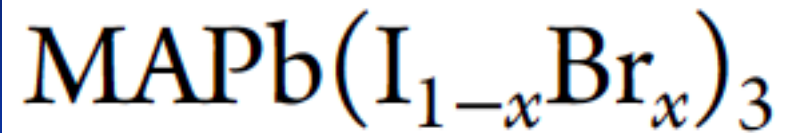


Nam Joong Jeon^{1†}, Jun Hong Noh^{1†}, Young Chan Kim¹, Woon Seok Yang¹, Seungchan Ryu¹ and Sang Il Seok^{1,2*}



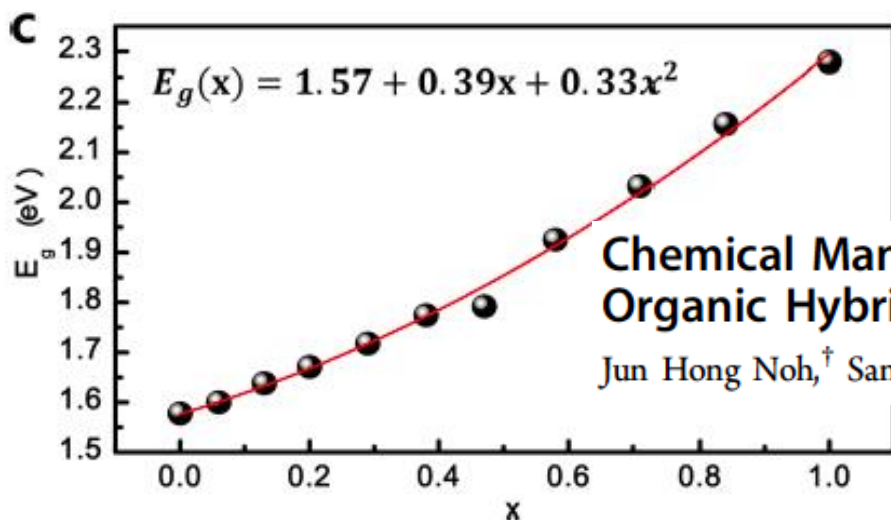
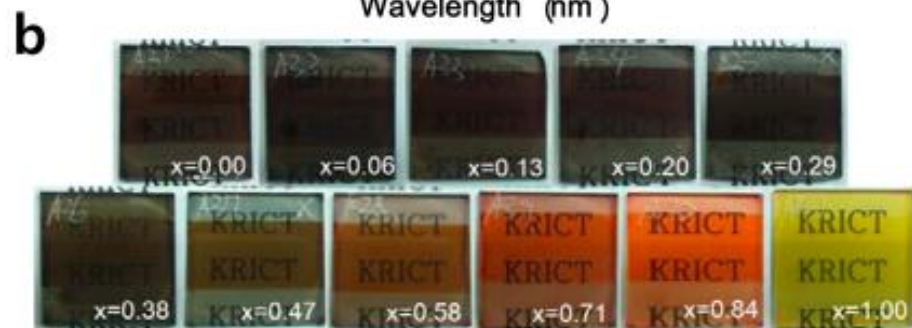
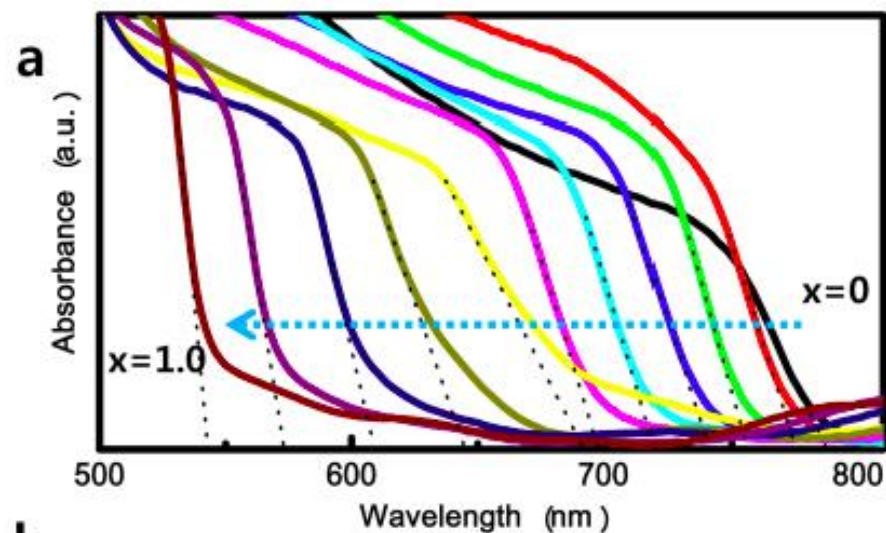
UNSW

Mixed compounds



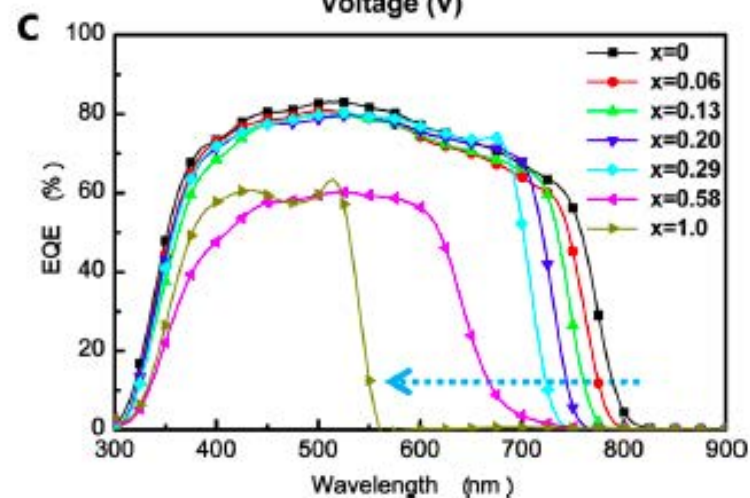
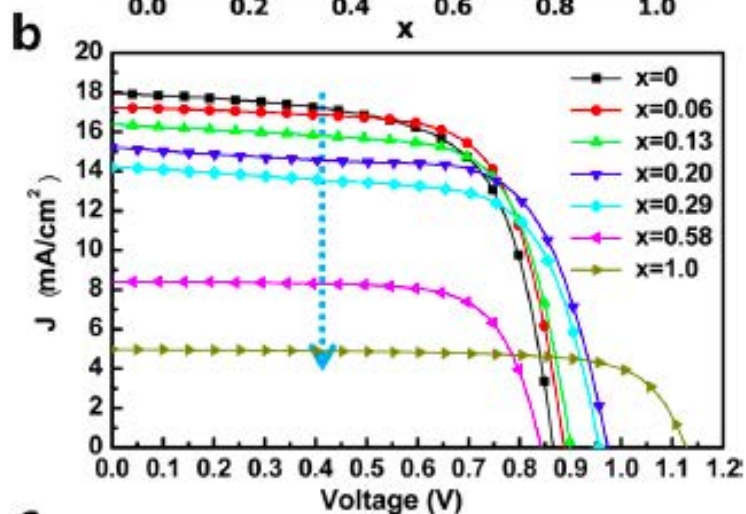
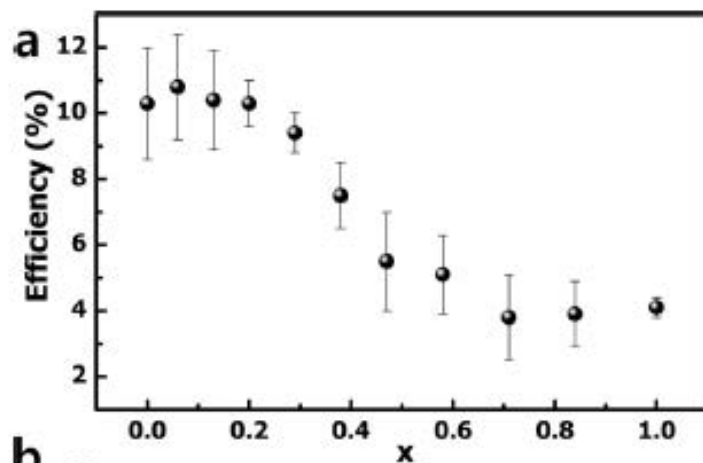
Chemical Management for Colorful, Efficient, and Stable Inorganic–Organic Hybrid Nanostructured Solar Cells

Jun Hong Noh,[†] Sang Hyuk Im,[‡] Jin Hyuck Heo,[†] Tarak N. Mandal,[†] and Sang Il Seok^{*,†,§}



Chemical Management
Organic Hybrid Nanost

Jun Hong Noh,[†] Sang Hyuk Im,[‡]

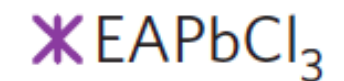
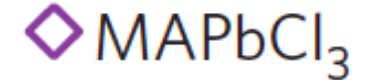
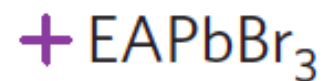
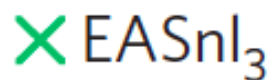
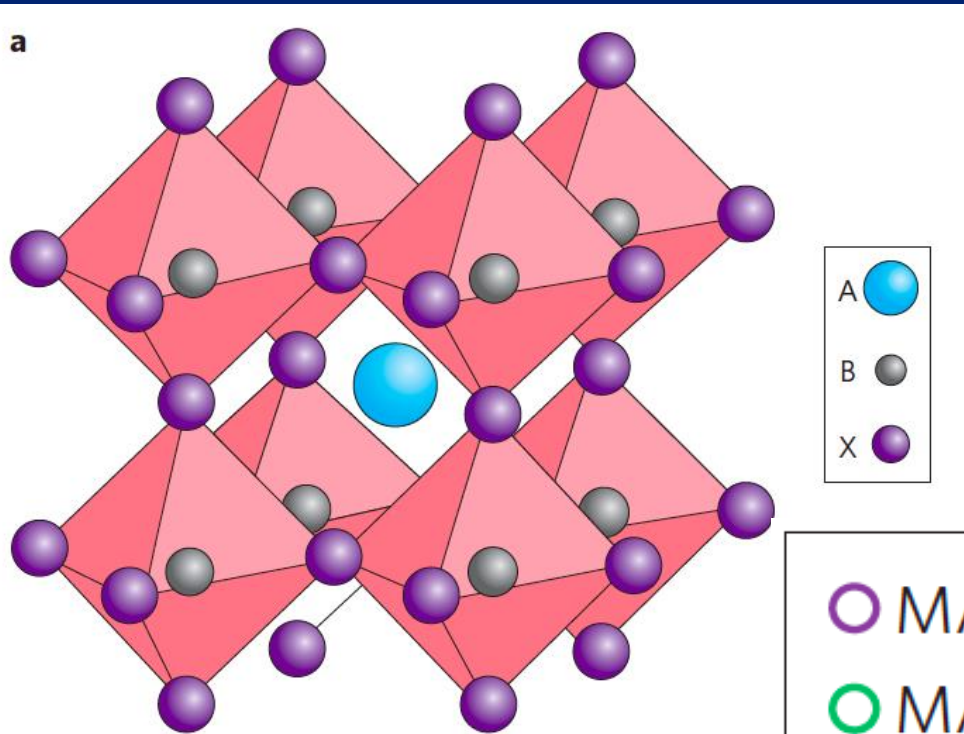


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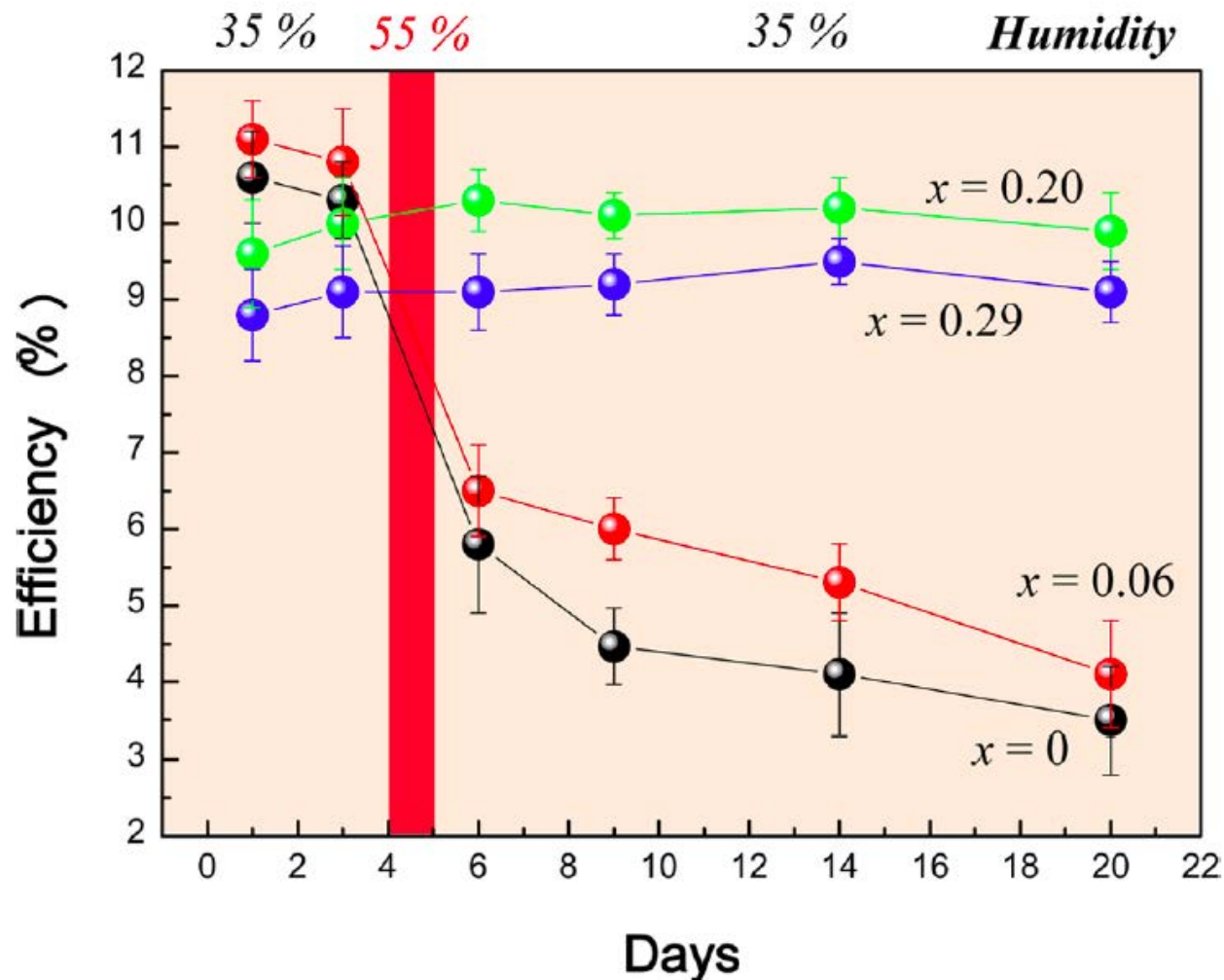
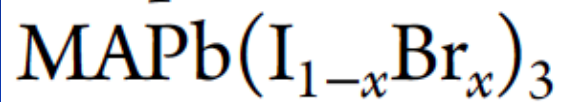
Mixed compounds





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Stability

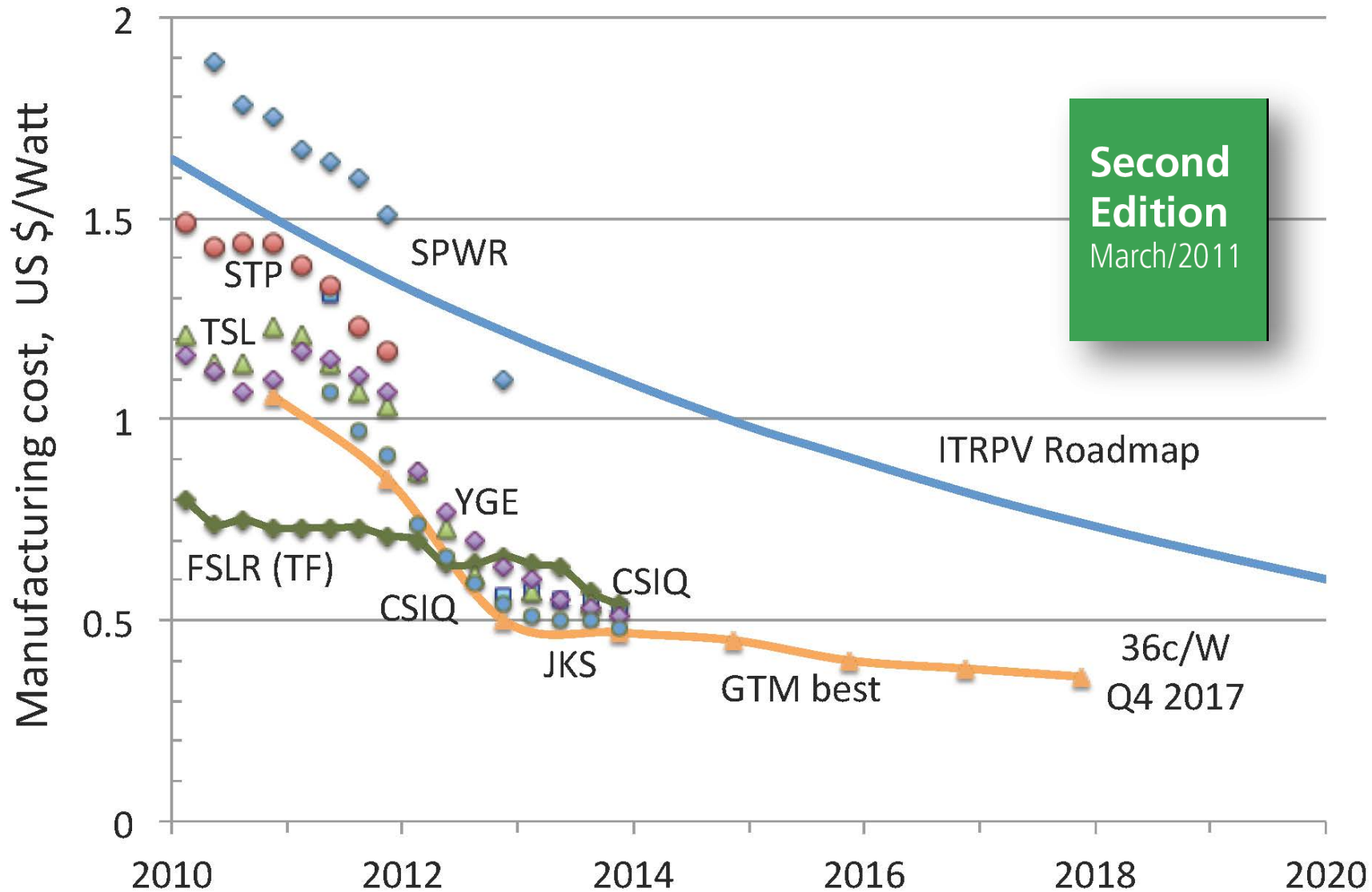


UNSW

Manufacturing costs

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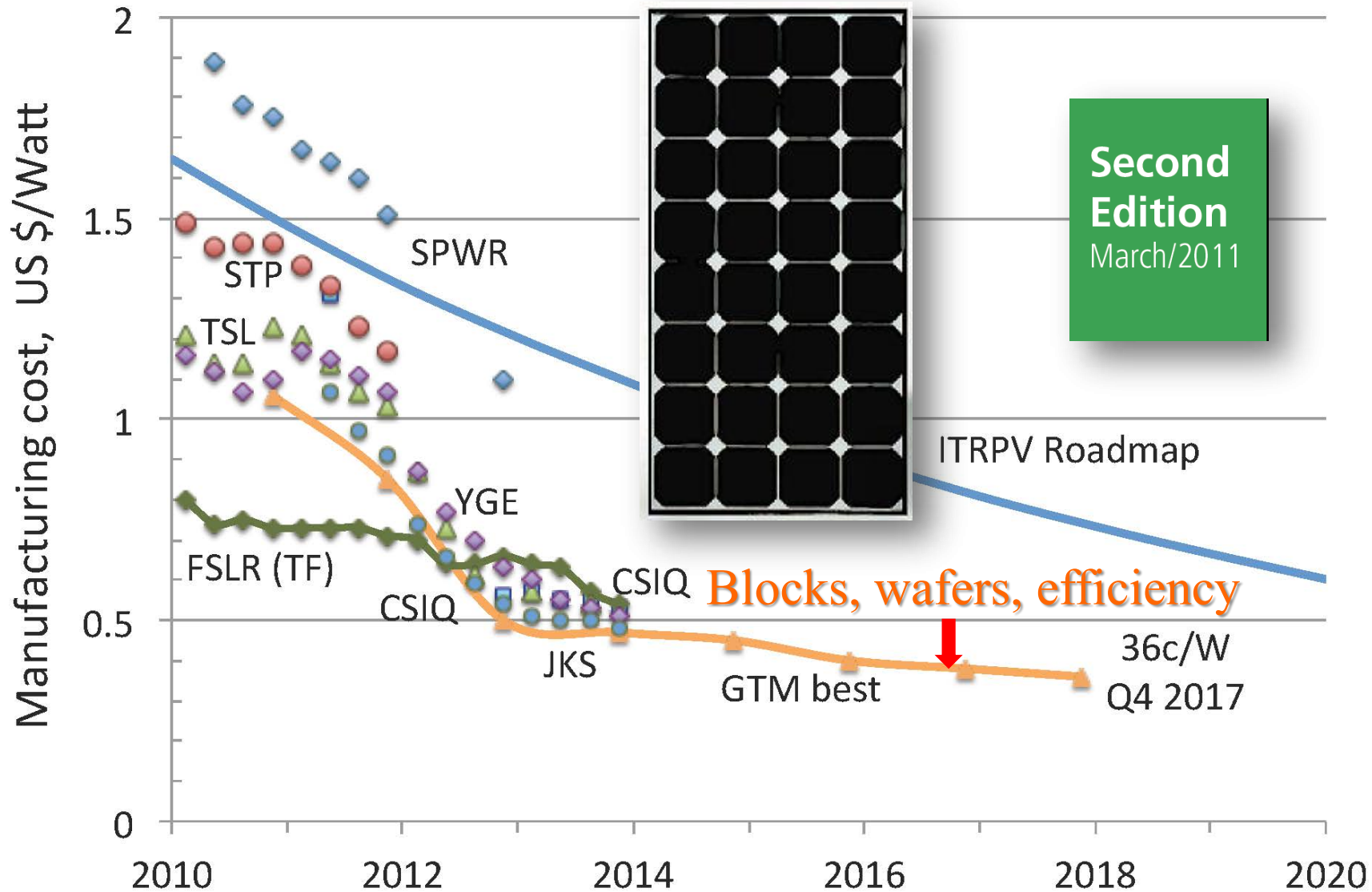
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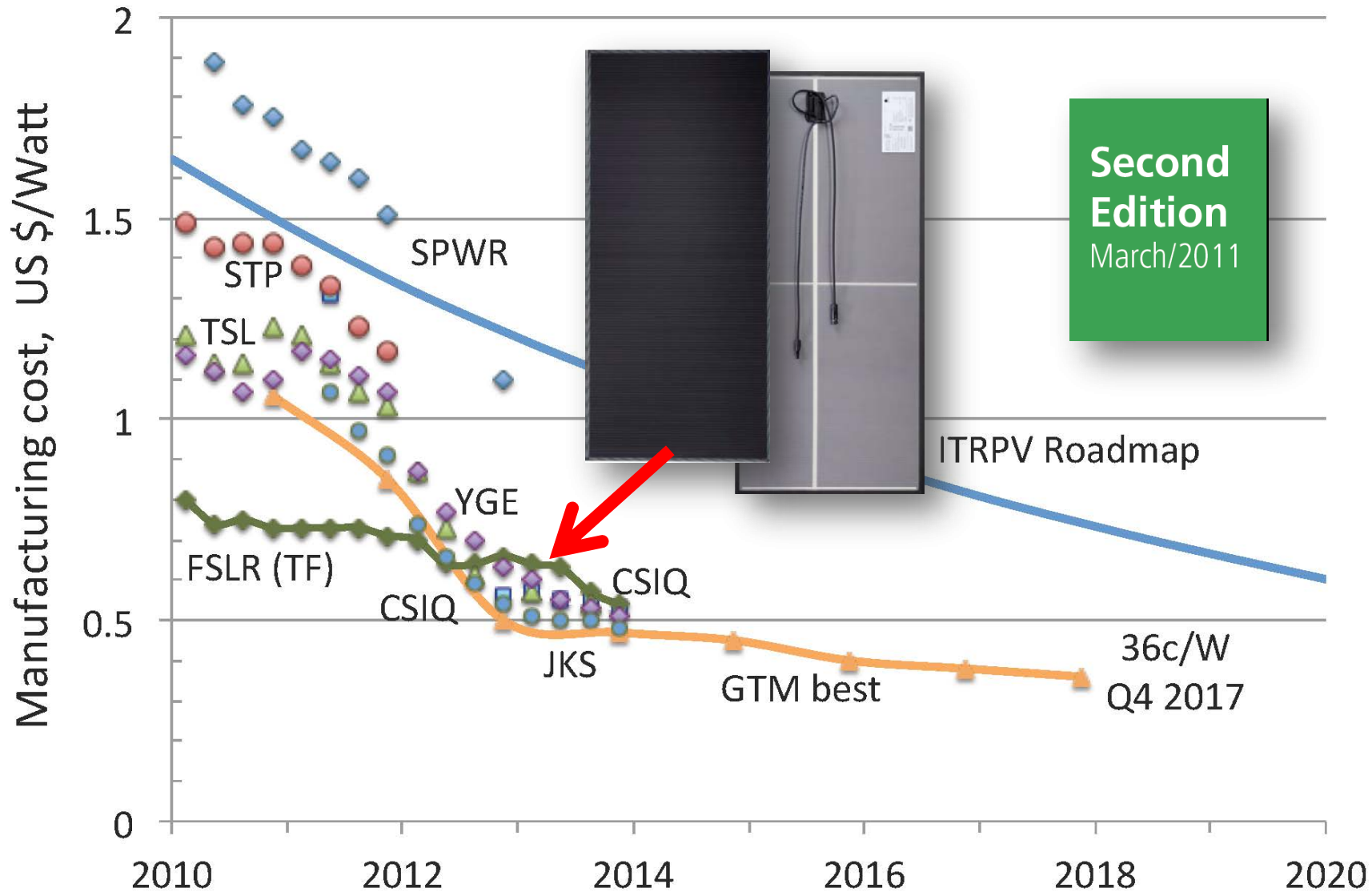
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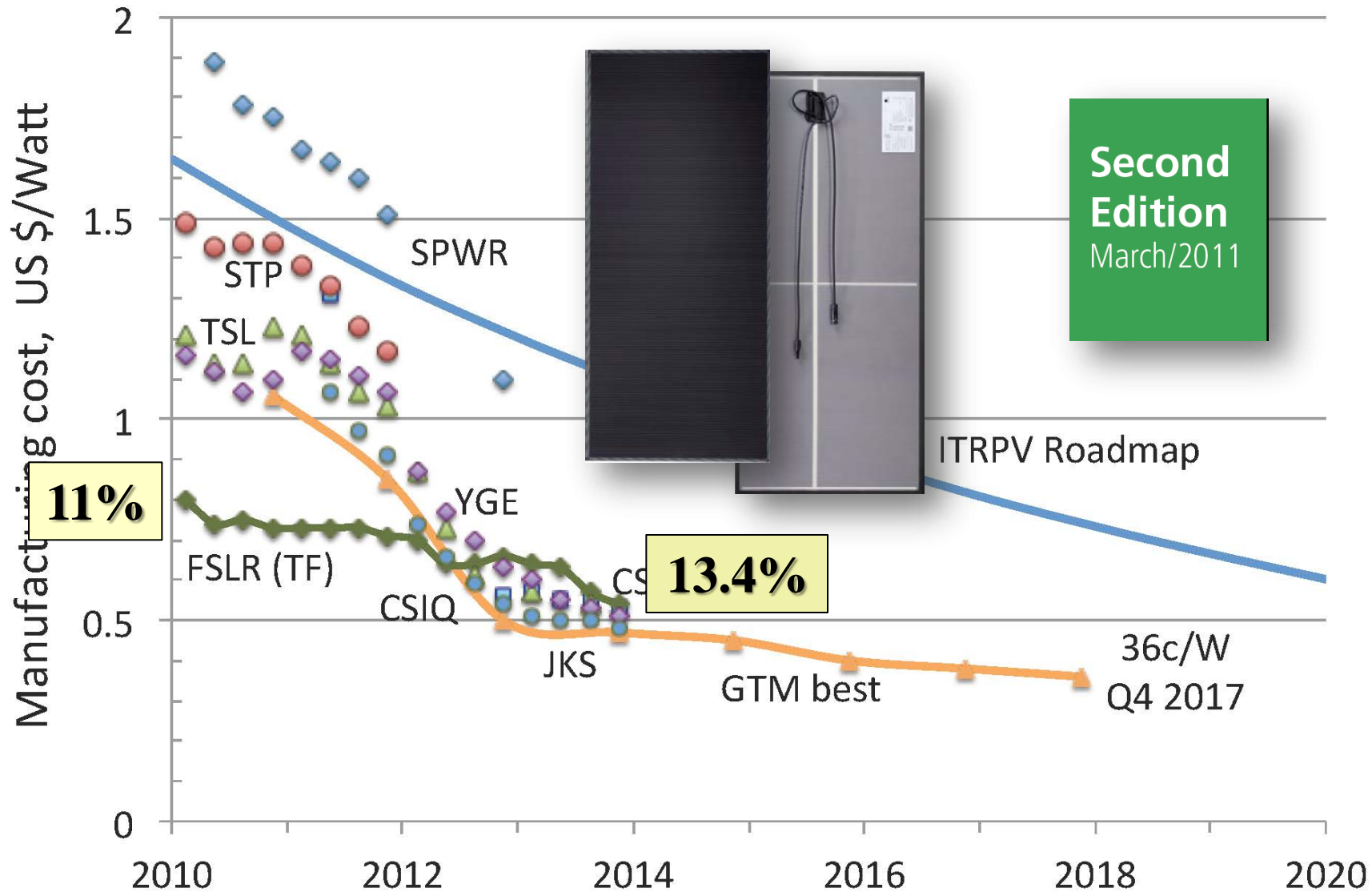
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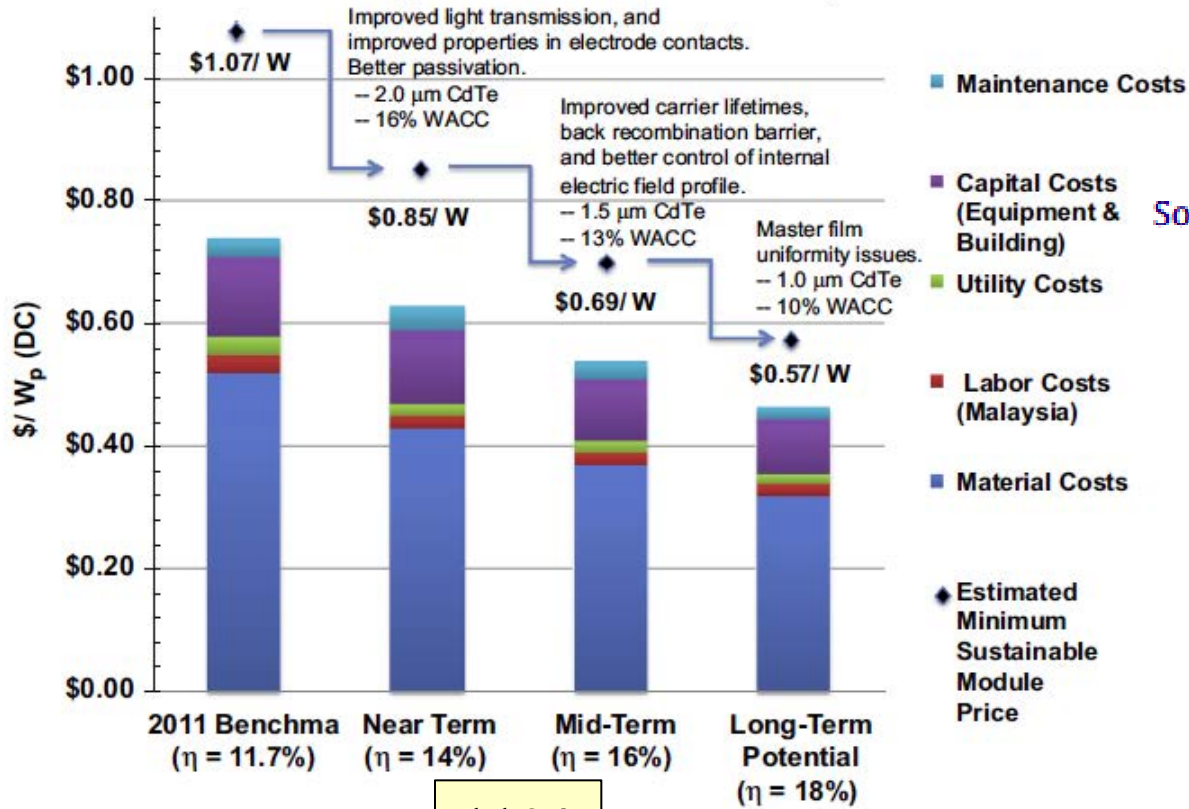
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Modeled CdTe Module Manufacturing Costs Under Dynamic Efficiency, CdTe Thickness, and WACC Inputs



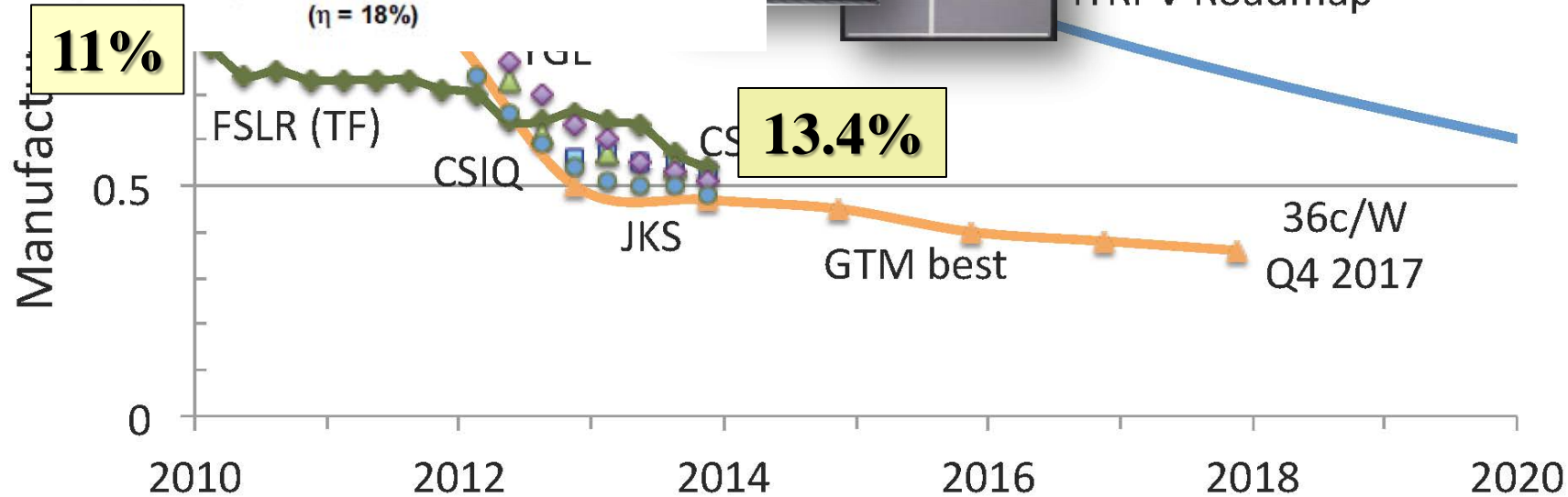
Manufacturing costs

Solar Energy Materials & Solar Cells 115 (2013) 199-212



Second Edition
 March/2011

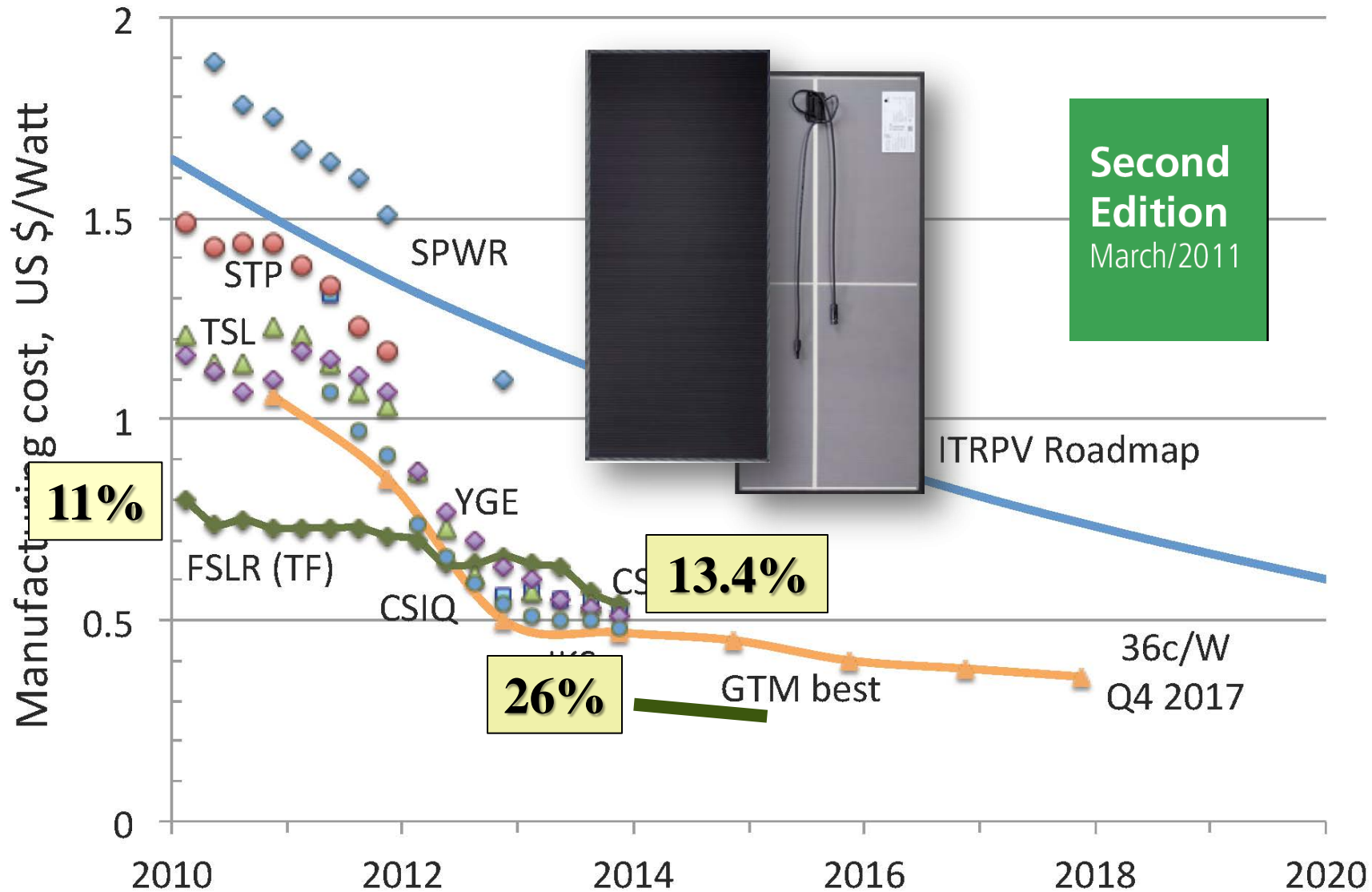
ITRPV Roadmap



Manufacturing costs

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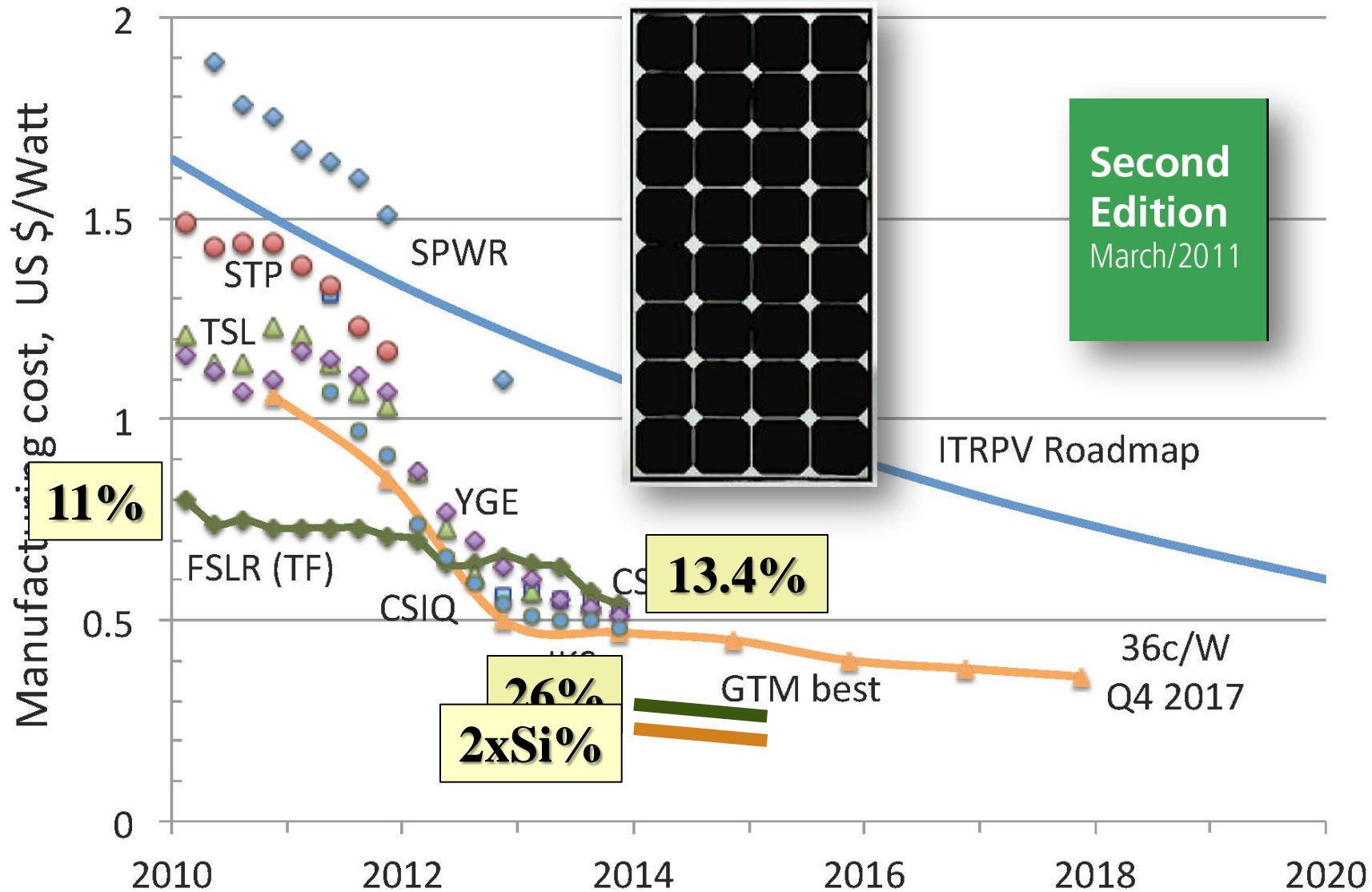
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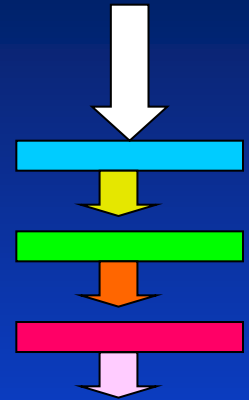
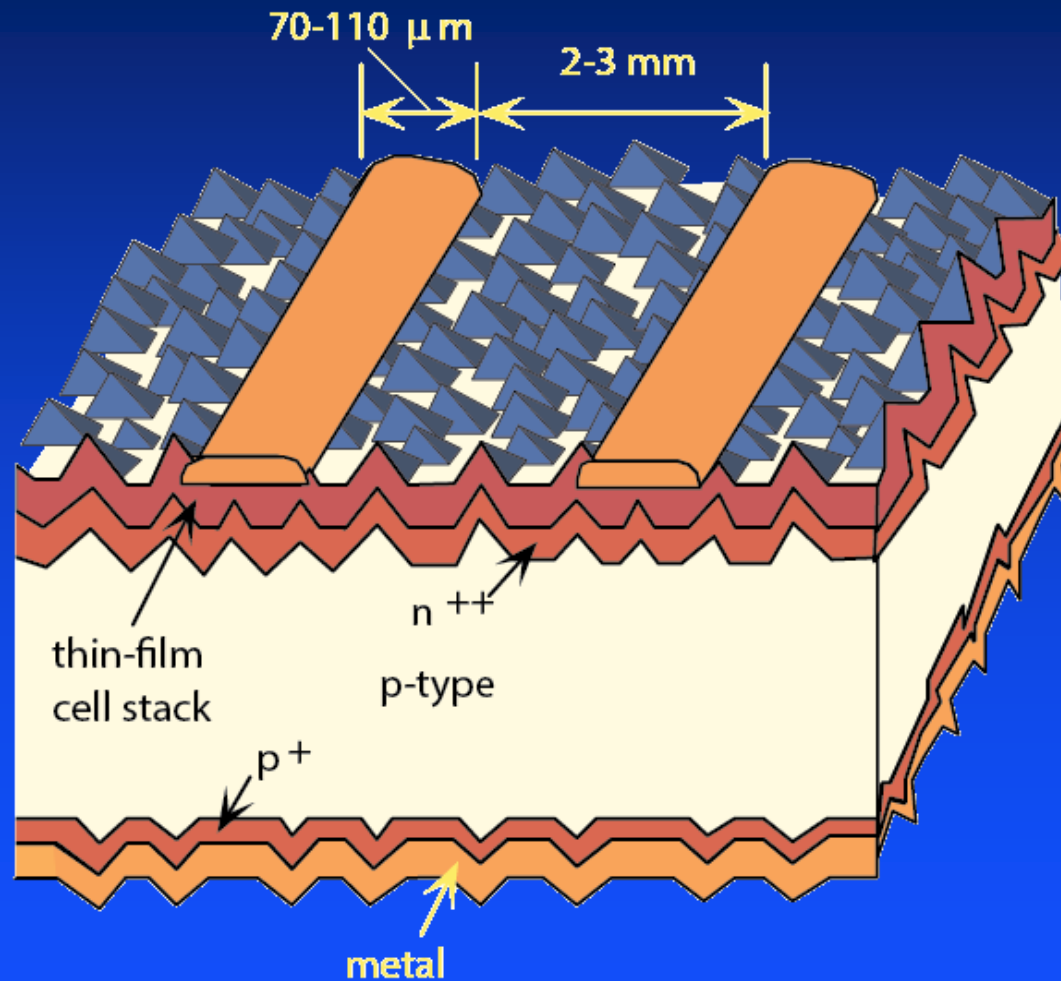
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Si wafer-based stack: ultimate solution?

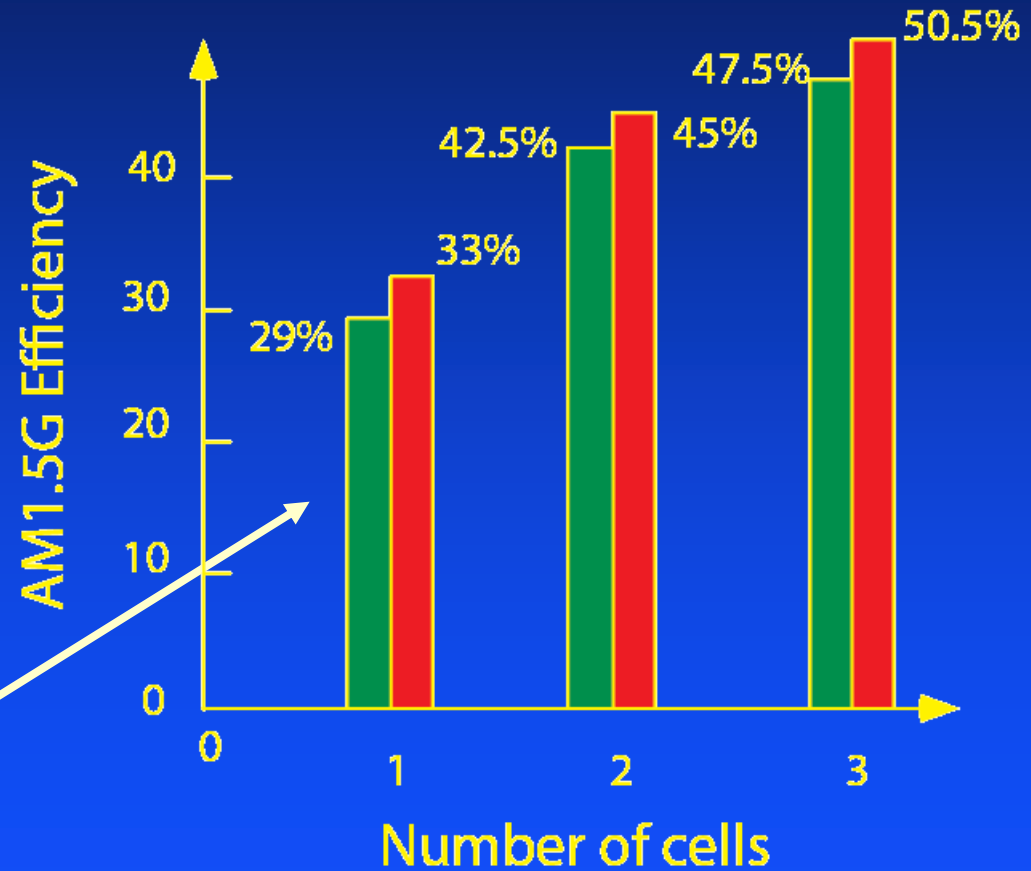
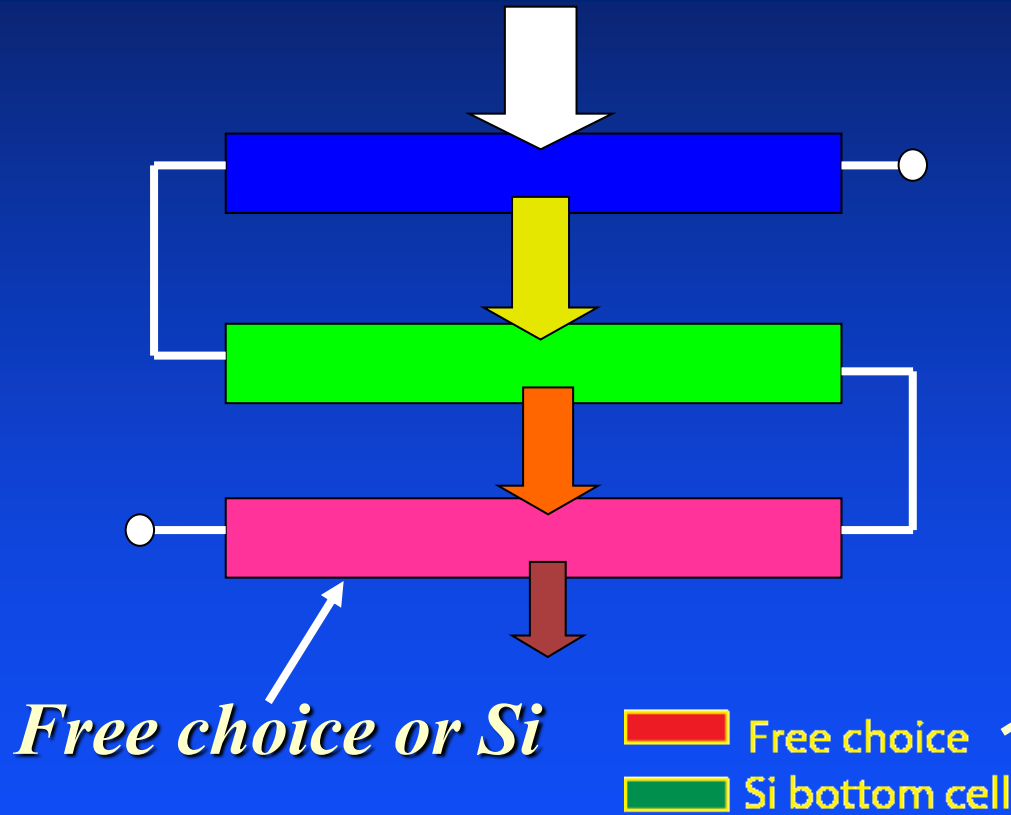
more sophisticated "active" AR coat?





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c-Si tandem

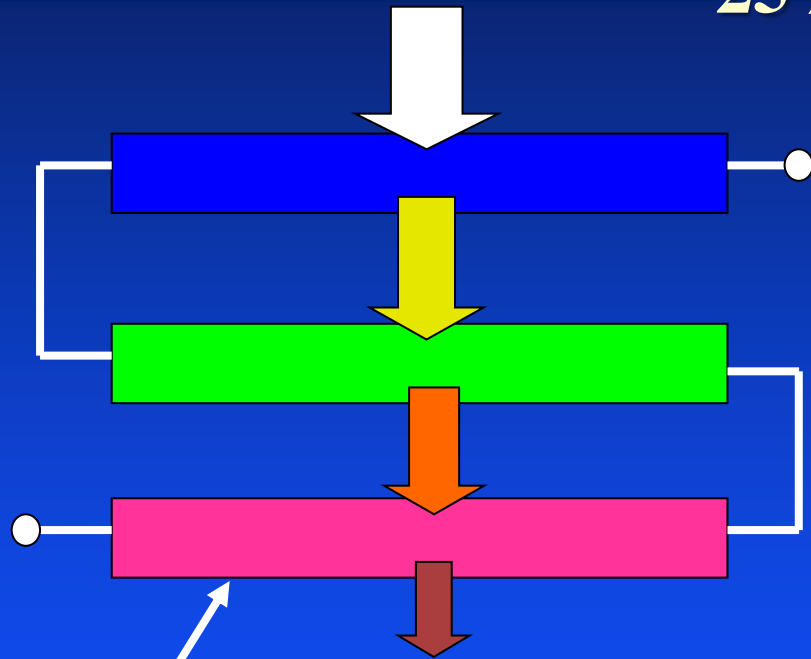




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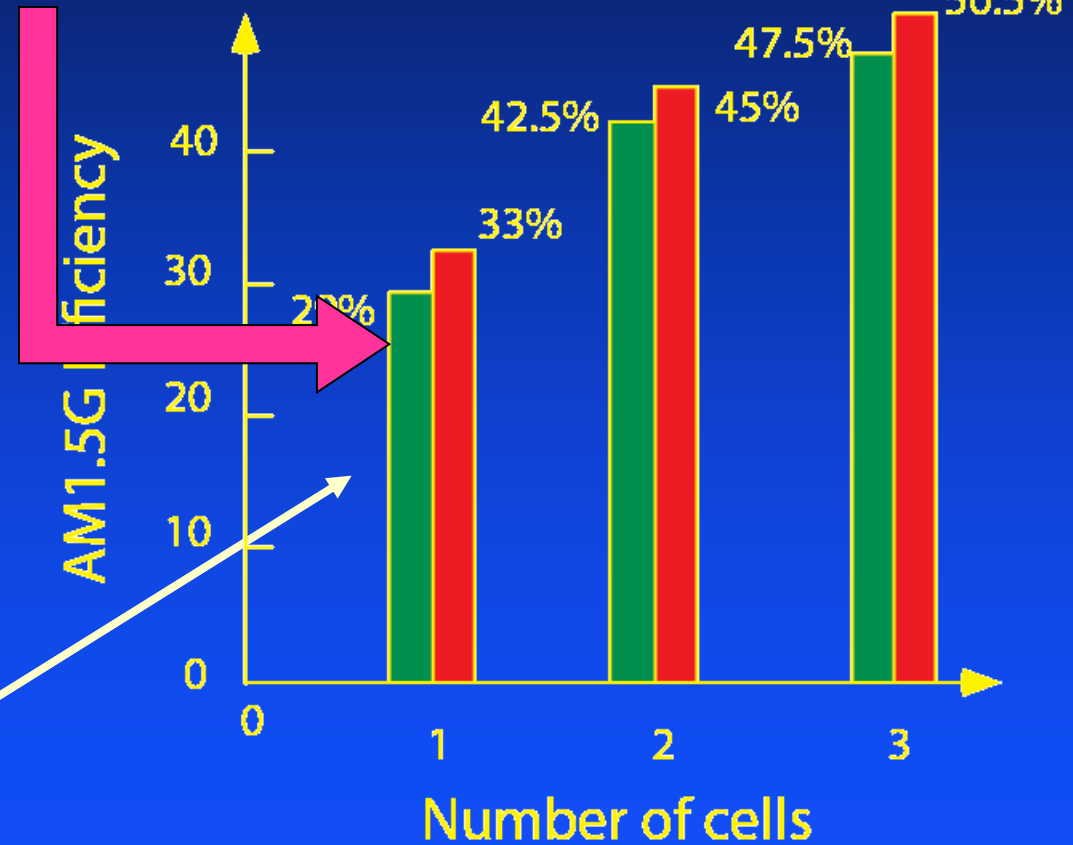
c-Si tandem

25% UNSW



Free choice or Si

- Free choice
- Si bottom cell

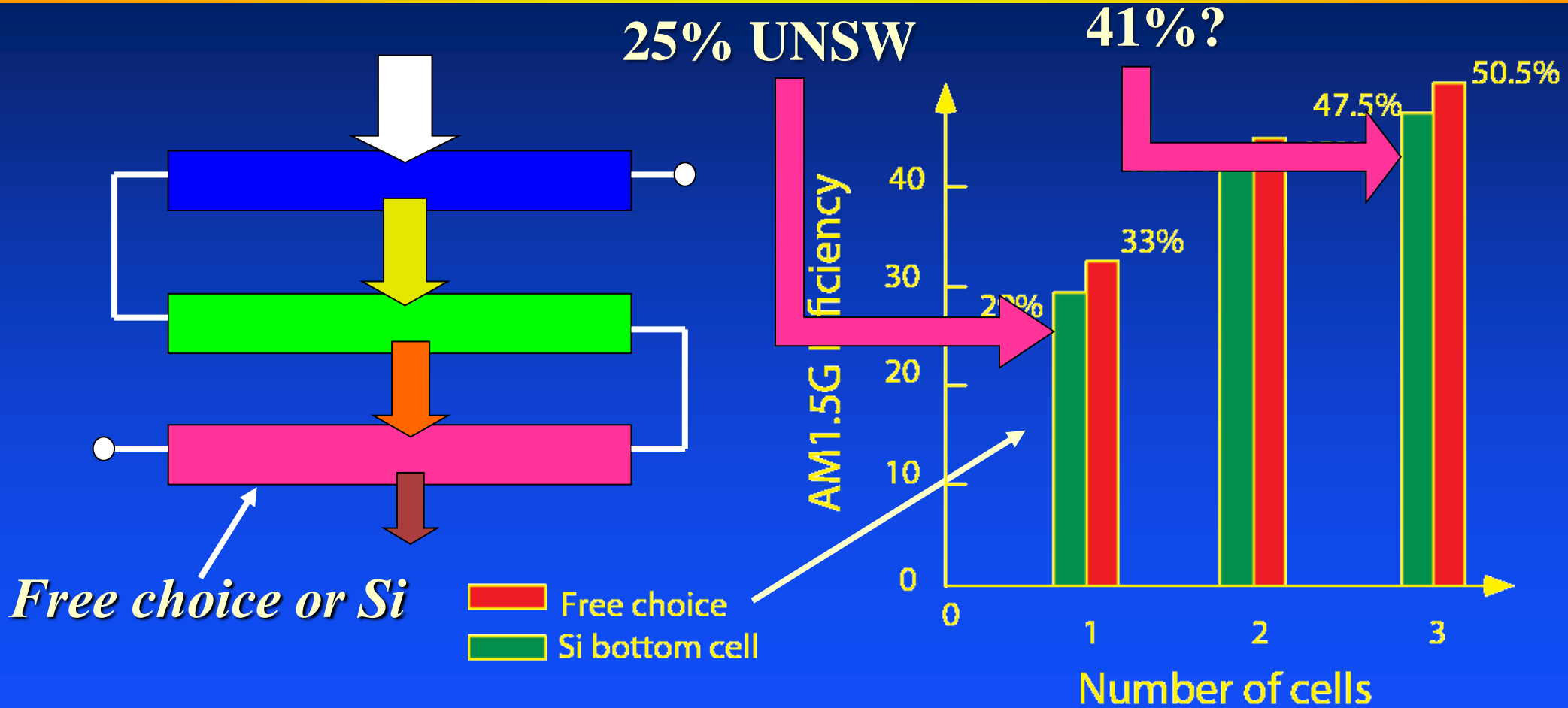


UNSW



ACAP

c-Si tandem

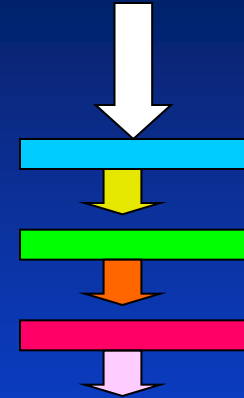
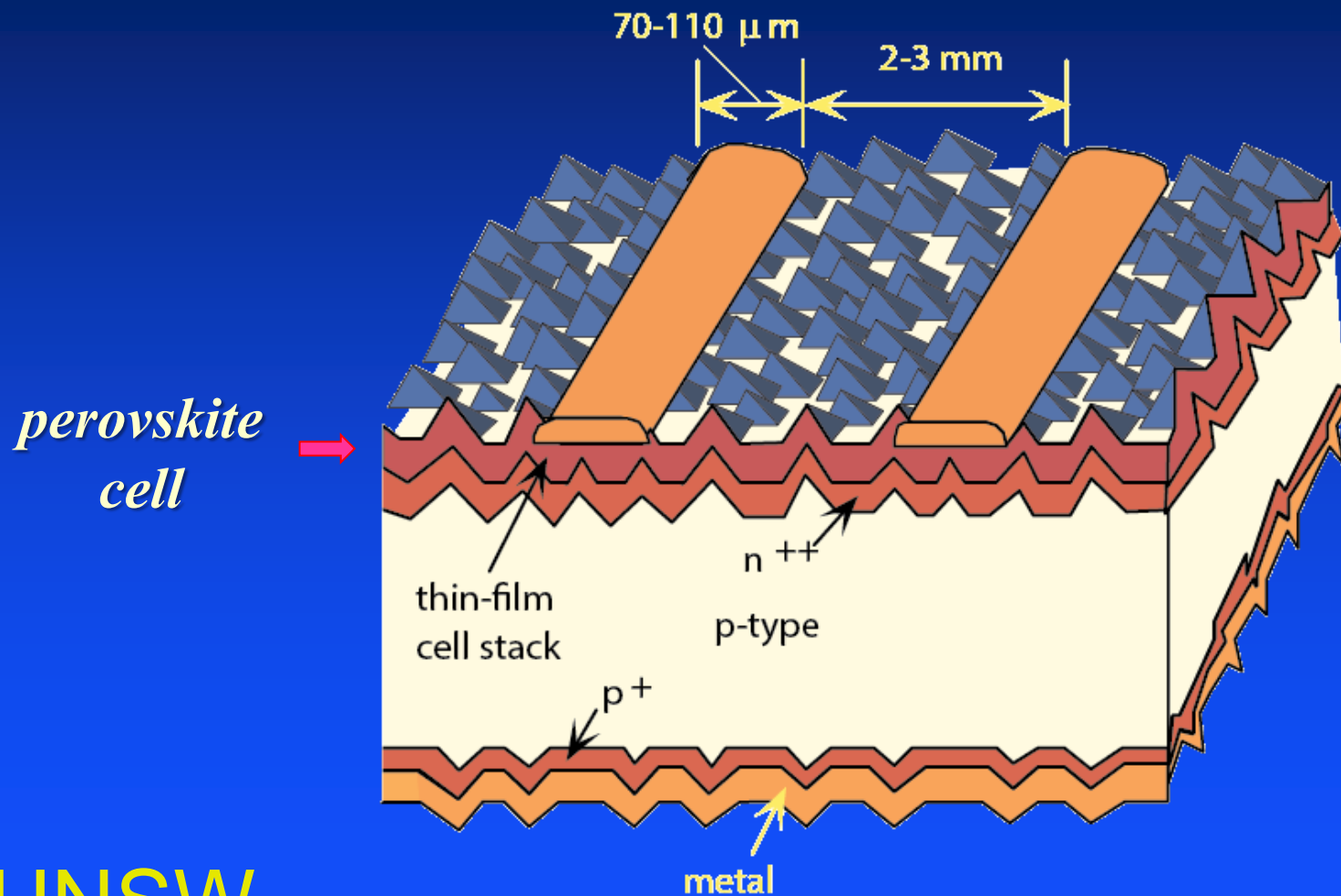


UNSW



ACAP

Perovskite on Si?

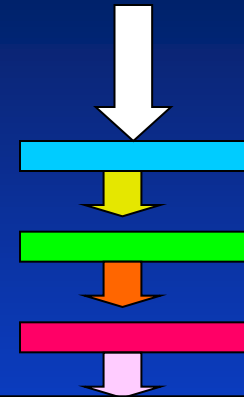
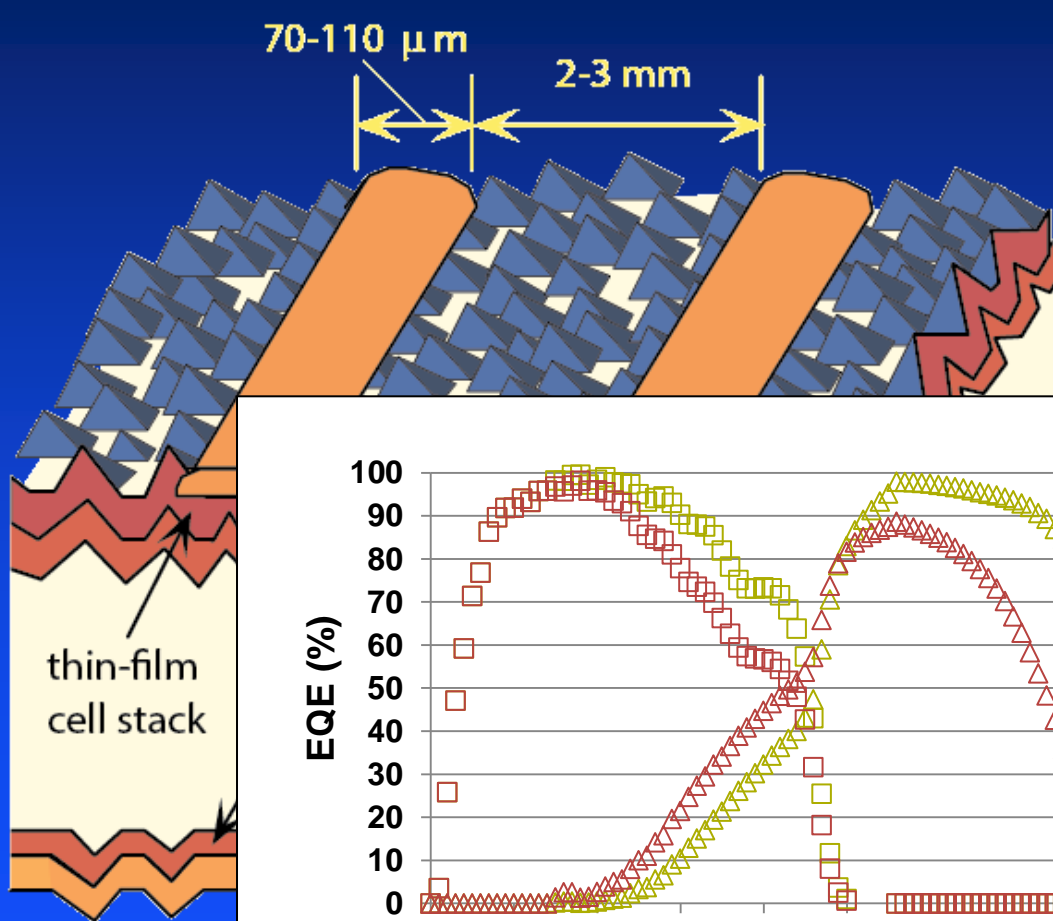


UNSW



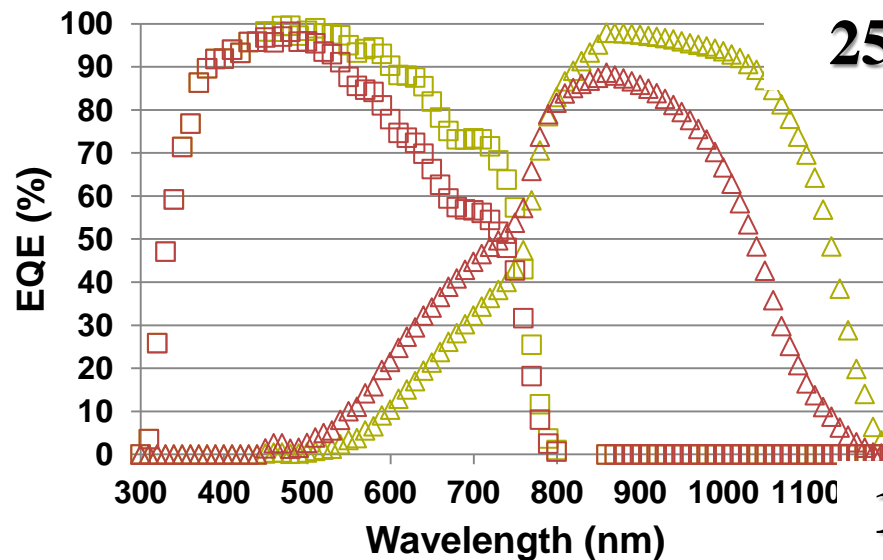
Perovskite on Si?

70-110 μm
2-3 mm



perovskite cell

thin-film cell stack



25% → 29.5%

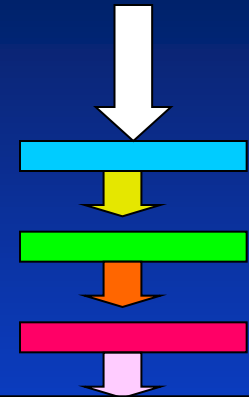
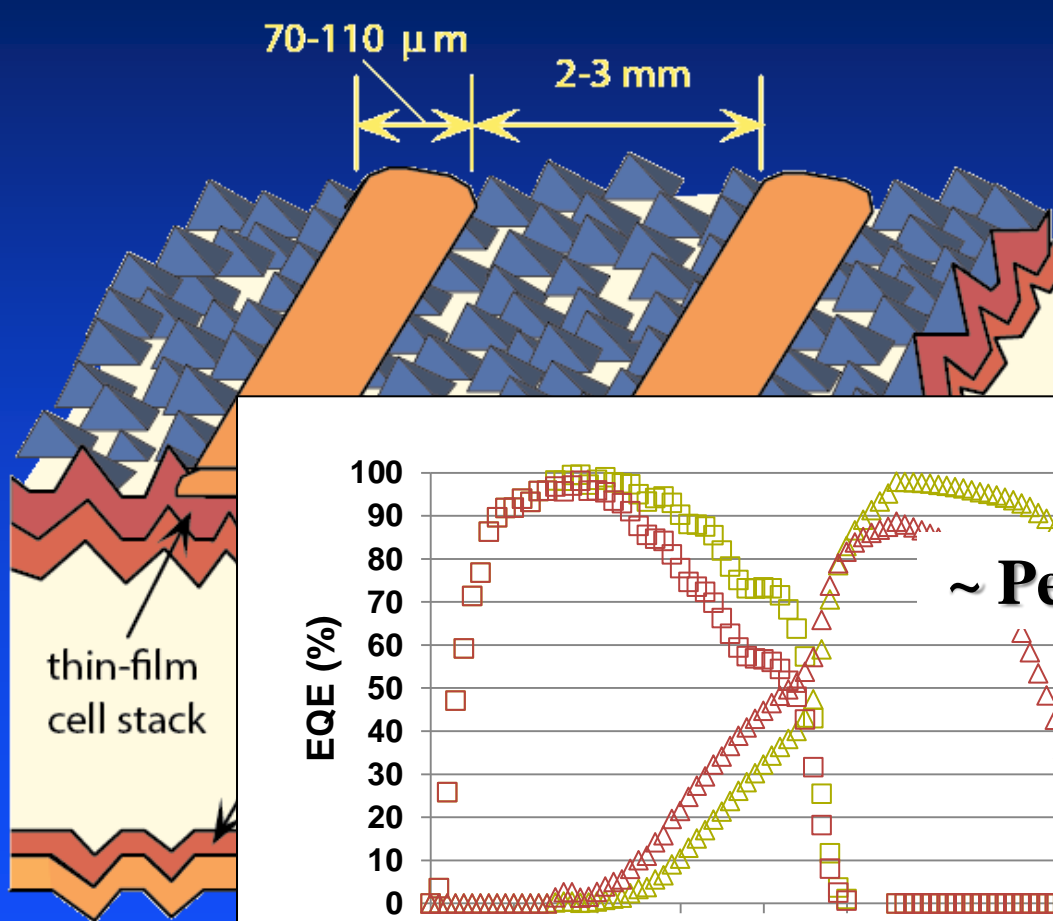
- Perovskite Top Cell on Si PERL
- △ Si PERL Bottom Cell
- Perovskite Top Cell on Si SP
- △ Si SP Bottom Cell

17.9% perovskite



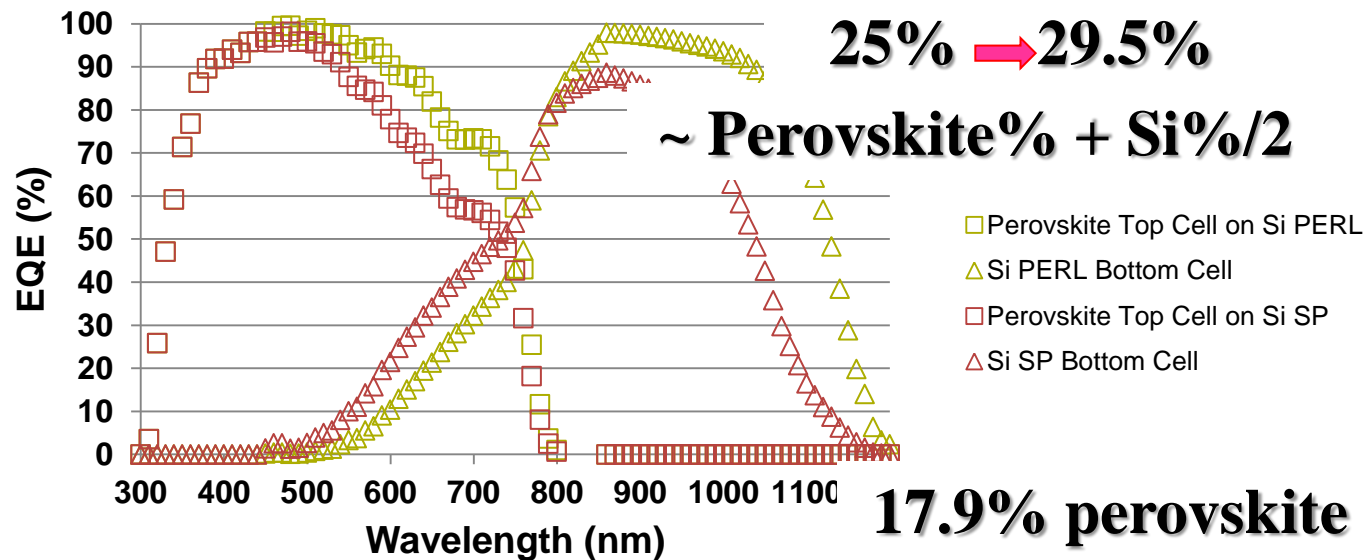
Perovskite on Si?

70-110 μm 2-3 mm



perovskite cell

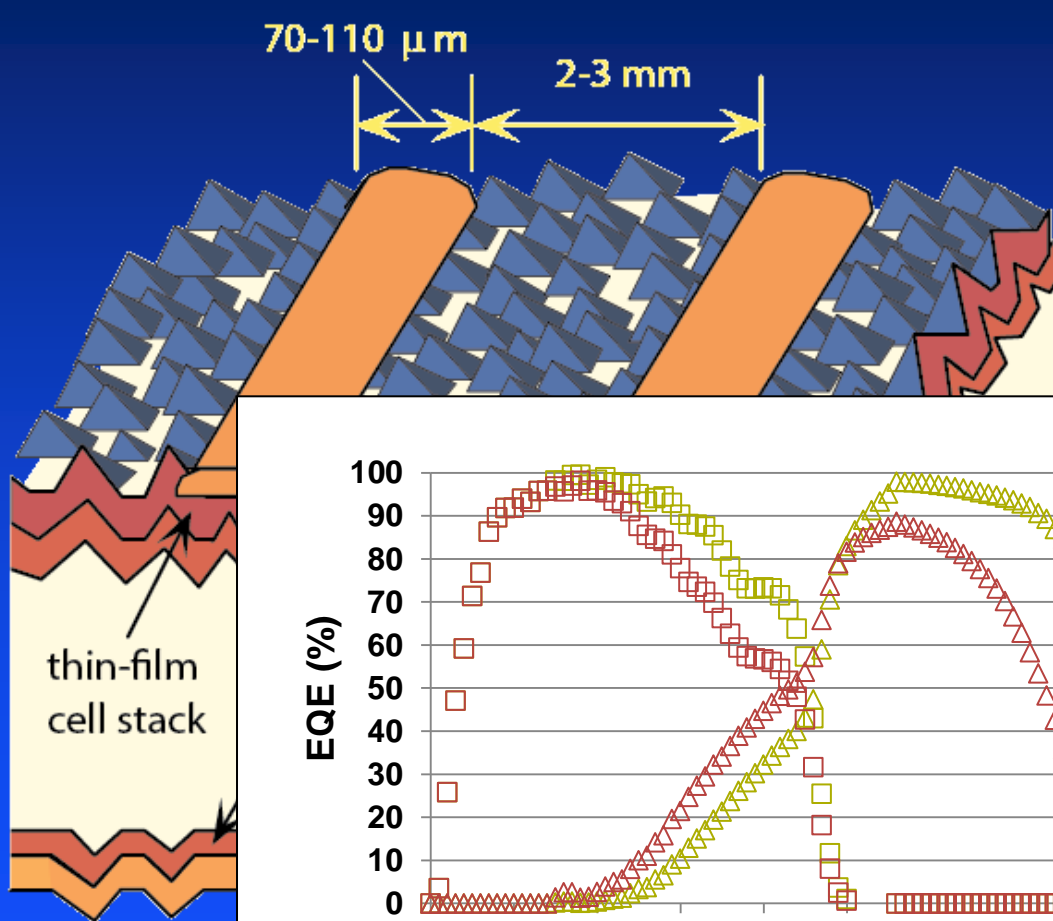
thin-film cell stack





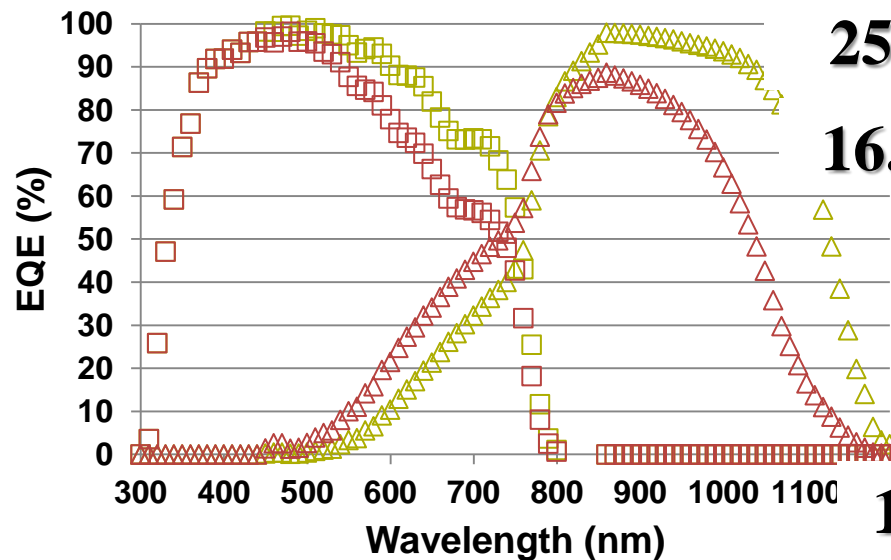
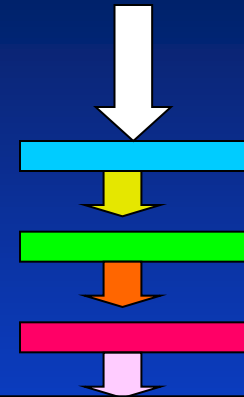
Perovskite on Si?

70-110 μm
2-3 mm



perovskite cell

thin-film cell stack



25% → 29.5%

16.5% → 25%

- Perovskite Top Cell on Si PERL
- △ Si PERL Bottom Cell
- Perovskite Top Cell on Si SP
- △ Si SP Bottom Cell

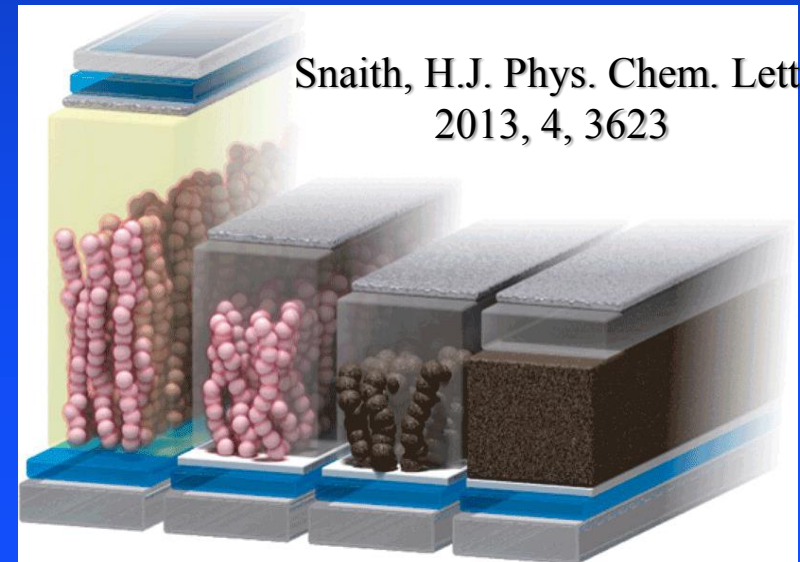
17.9% perovskite



Summary

. *Exciting time for perovskites*

UNSW



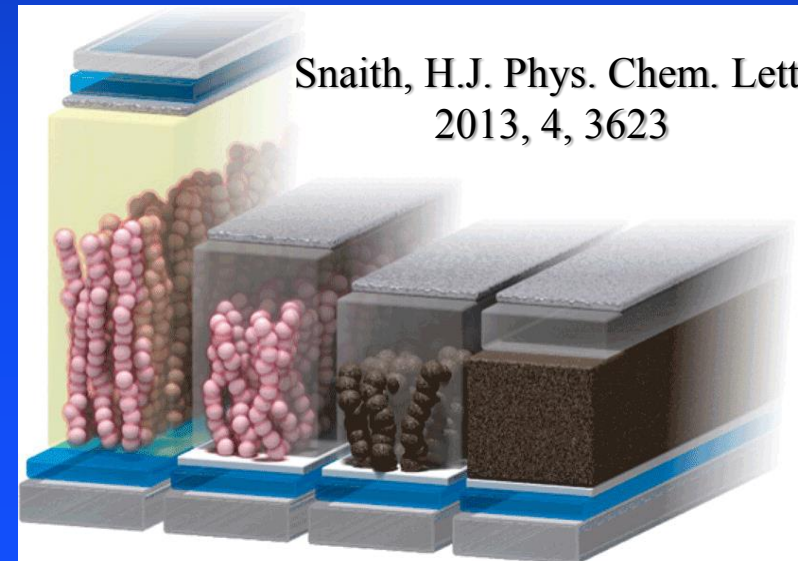


ACAP

Summary

- . *Exciting time for perovskites*
- . *Competitive advantages?:*

UNSW



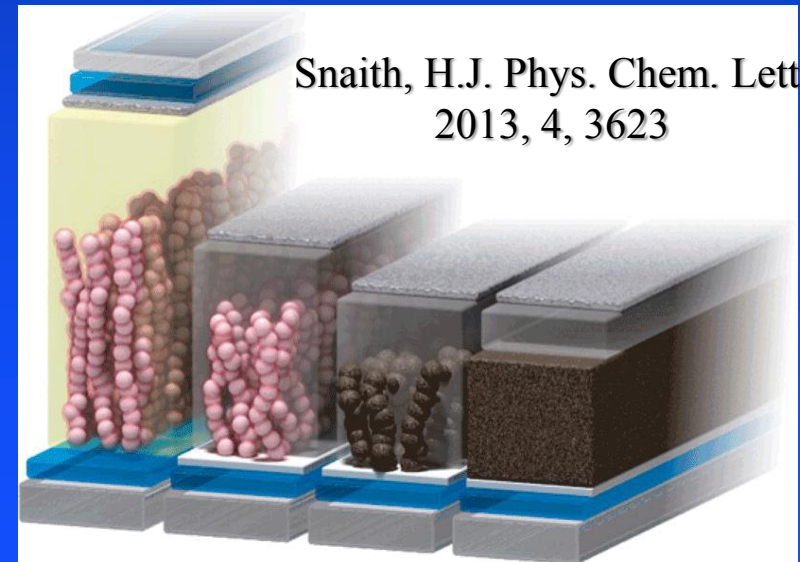


ACAP

Summary

- . *Exciting time for perovskites*
- . *Competitive advantages?:*
 - . *Low cost fabrication?*

UNSW

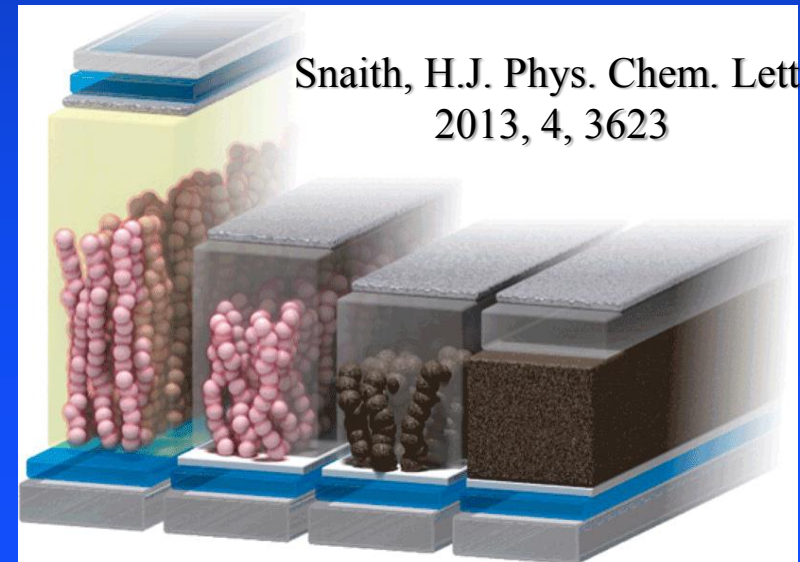




ACAP

Summary

- . *Exciting time for perovskites*
- . *Competitive advantages?:*
 - . *Low cost fabrication?*
 - . *Transparent and/or flexible product?*

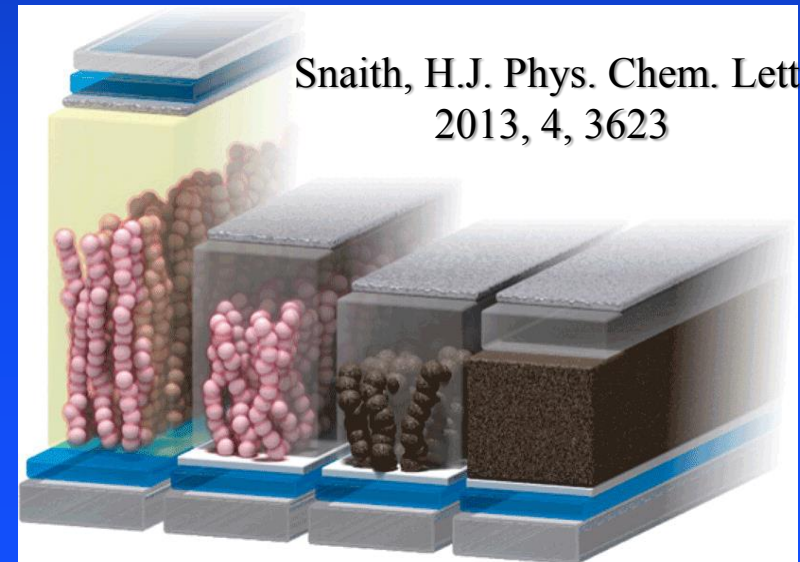




ACAP

Summary

- . *Exciting time for perovskites*
- . *Competitive advantages?:*
 - . *Low cost fabrication?*
 - . *Transparent and/or flexible product?*
 - . *Ability to form tandems?*





ACAP

Summary

- . *Exciting time for perovskites*
- . *Competitive advantages?:*
 - . *Low cost fabrication?*
 - . *Transparent and/or flexible product?*
 - . *Ability to form tandems?*
- . *Moisture sensitivity and toxicity of Pb*

